University of Debrecen Faculty of Science and Technology Institute of Physics

# PHYSICIST MSC PROGRAM

2023

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## **DEAN'S WELCOME**

Welcome to the Faculty of Science and Technology!

This is an exciting time for you, and I encourage you to take advantage of all that the Faculty of Science and Technology UD offers you during your bachelor's or master's studies. I hope that your time here will be both academically productive and personally rewarding

Being a regional centre for research, development and innovation, our Faculty has always regarded training highly qualified professionals as a priority. Since the establishment of the Faculty in 1949, we have traditionally been teaching and working in all aspects of Science and have been preparing students for the challenges of teaching. Our internationally renowned research teams guarantee that all students gain a high quality of expertise and knowledge. Students can also take part in research and development work, guided by professors with vast international experience.

While proud of our traditions, we seek continuous improvement, keeping in tune with the challenges of the modern age. To meet our region's demand for professionals, we offer engineering courses with a strong scientific basis, thus expanding our training spectrum in the field of technology. Recently, we successfully re-introduced dual training programmes in our constantly evolving engineering courses.

We are committed to providing our students with valuable knowledge and professional work experience, so that they can enter the job market with competitive degrees. To ensure this, we maintain a close relationship with the most important companies in our extended region. The basis for our network of industrial relationships are in our off-site departments at various different companies, through which market participants - future employers - are also included in the development and training of our students.

Prof. dr. Ferenc Kun Dean

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## **UNIVERSITY OF DEBRECEN**

Date of foundation: 1912 Hungarian Royal University of Sciences, 2000 University of Debrecen

**Legal predecessors**: Debrecen University of Agricultural Sciences; Debrecen Medical University; Wargha István College of Education, Hajdúböszörmény; Kossuth Lajos University of Arts and Sciences

Legal status of the University of Debrecen: state university

Founder of the University of Debrecen: Hungarian State Parliament

Supervisory body of the University of Debrecen: Ministry of Education

### Number of Faculties at the University of Debrecen: 13

Faculty of Agricultural and Food Sciences and Environmental Management Faculty of Child and Special Needs Education Faculty of Dentistry Faculty of Economics and Business Faculty of Engineering Faculty of Health Faculty of Humanities Faculty of Informatics Faculty of Law Faculty of Medicine Faculty of Music Faculty of Pharmacy Faculty of Science and Technology

## Number of students at the University of Debrecen: 29,777

## Full time teachers of the University of Debrecen: 1,587

203 full university professors and 1,249 lecturers with a PhD.

## FACULTY OF SCIENCE AND TECHNOLOGY

The Faculty of Science and Technology is currently one of the largest faculties of the University of Debrecen with about 2,500 students and more than 200 staff members. The Faculty has got 6 institutes: Institute of Biology and Ecology, Institute of Biotechnology, Institute of Chemistry, Institute of Earth Sciences, Institute of Physics and Institute of Mathematics. The Faculty has a very wide scope of education dominated by science and technology (12 Bachelor programs and 14 Master programs), additionally it has a significant variety of teachers' training programs. Our teaching activities are based on a strong academic and industrial background, where highly qualified teachers with a scientific degree involve student in research and development projects as part of their curriculum. We are proud of our scientific excellence and of the application-oriented teaching programs with a strong industrial support. The number of international students of our faculty is continuously growing (currently ~790 students). The attractiveness of our education is indicated by the popularity of the Faculty in terms of incoming Erasmus students, as well.

#### THE ORGANIZATIONAL STRUCTURE OF THE FACULTY

Dean: Prof. Dr. Ferenc Kun, Full Professor E-mail: <u>ttkdekan@science.unideb.hu</u>

Vice Dean for Educational Affairs: Prof. Dr. Gábor Kozma, Full Professor E-mail: <u>kozma.gabor@science.unideb.hu</u>

Vice Dean for Scientific Affairs: Prof. Dr. Sándor Kéki, Full Professor E-mail: <u>keki.sandor@science.unideb.hu</u>

Consultant on External Relationships: Prof. Dr. Attila Bérczes, Full Professor E-mail: <u>berczesa@science.unideb.hu</u>

Consultant on Talent Management Programme: Prof. dr. Tibor Magura, Full Professor E-mail: <u>magura.tibor@science.unideb.hu</u>

Dean's Office Head of Dean's Office: Mrs. Katalin Kozma-Tóth E-mail: toth.katalin@science.unideb.hu

English Program Officer: Mrs. Alexandra Csatáry Address: 4032 Egyetem tér 1., Chemistry Building, A/101, E-mail: <u>acsatary@science.unideb.hu</u>

# DEPARTMENTS OF INSTITUTE OF PHYSICS

## Department of Experimental Physics (home page: https://fizika.unideb.hu/en/departmentexperimental-physics-introduction) 4026 Debrecen, Bem tér 18/a,

Name	Position	E-mail	room
Mr. Prof. Dr. Zoltán	University Professor	zoltan.trocsanyi@science.unideb.hu	F21
Trócsányi, PhD, habil,			
DSc, Member of HAS			
Mr. Dr. István Nándori,	Associate Professor	nandori.istvan@science.unideb.hu	F020
PhD, habil			
Mr. Dr. Gyula Zilizi,	Associate Professor,	zilizi@science.unideb.hu	E116
PhD, habil	Head of Department		
Mr. Dr. István	Associate Professor	csarnovics.istvan@science.unideb.hu	E214
Csarnovics, PhD			
Ms. Dr. Judit Darai,	Associate Professor	darai@science.unideb.hu	E116
PhD, habil			
Mr. Dr. Sándor Egri,	Assistant Professor	egris@science.unideb.hu	E209
PhD			
Mr. Dr. László Oláh,	Assistant Professor	olah.laszlo@science.unideb.hu	E115
PhD			
Mr. Roland Szatmári	Assistant	szatmari.roland@science.unideb.hu	E209
Mr. Dr. Ádám Kardos,	Assistant Professor	kardos.adam@science.unideb.hu	E207
PhD			
Dr. Dezső Horváth,		Horvath.dezso@wigner.hu	F021
Dsc			

## **Department of Theoretical Physics** (home page: http://www.phys.unideb.hu/dtp/) **4026 Debrecen, Bem tér 18/b**

Name	Position	E-mail	room
Ms. Prof. Dr. Ágnes	University Professor,	vibok.agnes@science.unideb.hu	E2
Vibók,	Head of Department		
PhD, habil, DSc			
Mr. Prof. Dr. Ferenc	University Professor	ferenc.kun@science.unideb.hu	E1
Kun,			
PhD, habil, DSc,			
Member of HAS			
Mr. Dr. Sándor Nagy,	Associate Professor	Professor sandor.nagy@science.unideb.hu	
PhD, habil			
Mr. Dr. András Csehi,	Assistant Professor	csehi.andras@science.unideb.hu	E10
PhD			
Mr. Dr. Zsolt Schram,	Associate Professor	schram@unideb.hu	E4
PhD habil			
Mr. Dr. Gergő Pál,	Assistant Professor	gergo.pal@phys.unideb.hu	E3
PhD			
Mr. Dr. Péter Badankó,	Research Assistant	badanko.peter@gmail.com	E10
PhD			

## Department of Condensed Matter Physics (home page: http://lolka.phys.unideb.hu) 4026 Debrecen, Bem tér 18/b

Name	Position	E-mail	room
Mr. Prof. Dr. Zoltán	University Professor,	zoltan.erdelyi@science.unideb.hu	E8
Erdélyi, PhD, habil,	Head of Department		
DSc			
Mr. Dr. Lajos Daróczi,	Associate Professor	lajos.daroczi@science.unideb.hu	F9
PhD, habil			
Mr. Dr. Gábor Katona,	Assistant Professor	gabor.katona@science.unideb.hu	F2
PhD			
Mr. Dr. Csaba Cserháti,	Associate Professor	cserhati.csaba@science.unideb.hu	F10
PhD, habil			
Mr. János Tomán,	Assistant Lecturer	janos.toman@science.unideb.hu	F10
Mr. Dr. István Szabó,	Associate Professor	istvan.szabo@science.unideb.hu	F20
PhD, habil			
Mr. Dr. László Tóth,	Assistant Lecturer	toth.laszlo@science.unideb.hu	F2
PhD			
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Mr. Lajos Harasztosi	Teacher of	lajos.harasztosi@science.unideb.hu	F9
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## **Department of Electrical Engeneering** (home page: http://eed.science.unideb.hu) **4026 Debrecen, Bem tér 18/a**

Name	Position	E-mail	room
Mr. Prof. Dr. Gábor	University Professor,	battistig.gabor@science.unideb.hu	E114
Battistig, PhD, habil,	Head of Department		
DSc			
Mr. Dr. János Kósa,	Assistant Professor	kosa.janosarpad@science.unideb.hu	U5/A
PhD			
Mr. Dr. Sándor Misák,	College Associate	misak@science.unideb.hu	E214
PhD	Professor		
Mr. Árpád Rácz	Assistant Lecturer	racz.arpad@science.unideb.hu	U5/A
Ms. Dr. Réka	Assistant Professor	trencsenyi.reka@science.unideb.hu	U3
Trencsényi, PhD			
Ms. Tünde Anna	Master Teacher	bne.toth.tunde@science.unideb.hu	U6
Baloghné Tóth			
Mrs. Dr. Kósáné	Teacher of	kalave.eniko@science.unideb.hu	E205
Kalavé Enikő	engineering		
Mr. Zsolt Markovics	Teacher of	markovics.zsolt@science.unideb.hu	U6
	engineering		
Mr. Zsolt Szabó	Teacher of	szabo.zsolt@science.unideb.hu	U4
	engineering		
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	engineering		
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DSc.			
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	engineeering		

Ms. Rebeka kiss	Teacher of	kiss.rebeka@science.unideb.hu	
	engierreing		

## Department of Environmental Physics (home page: http://w3.atomki.hu/deat/) 4026 Debrecen, Bem tér 18/c

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PhD, habil	head of department		
Dr. Eszter Baradács,	Assistant Professor	baradacs@science.unideb.hu	
PhD			
Dr. Zoltán Papp,	Associate Professor	zpapp@science.unideb.hu	
PhD, habil			

## ACADEMIC CALENDAR

General structure of the academic semester (2 semesters/year):

Study pariod	1 <sup>st</sup> week	Registration*	1 week
Study period	$2^{nd} - 15^{th}$ week	Teaching period	14 weeks
Exam period	directly after the study period	Exams	7 weeks

\*Usually, registration is scheduled for the first week of September in the fall semester, and for the first week of February in the spring semester.

For further information please check the following link:

https://www.edu.unideb.hu/tartalom/downloads/University\_Calendars\_2023\_24/University\_calendar\_2023-2024-Faculty\_of\_Science\_and\_Technology.pdf

## THE PHYSICIST MASTER PROGRAM

## **Information about the Program**

Name of MSc Program:	Physicist MSc Program
Specialization available:	
Field, branch:	Science
Qualification:	Physicist
Mode of attendance:	Full-time
Faculty, Institute:	Faculty of Science and Technology
	Institute of Physics
Program coordinator:	Prof. Dr. Ágnes Vibók, University Professor
Duration:	4 semesters
ECTS Credits:	120

## **Objectives of the MSc program:**

The objective of the programme is to train physicist, who is able to realize physical principles in natural phenomena, to perform their experimental investigation according to scientific standards, and to obtain the theoretical understanding. The training enables her/him to develop and operate industrial, IT, and measuring systems related to physical laws and high technology processes. The student is able to continuously broaden her/his knowledge and has the aptitude for continuing her/his studies in the framework of doctoral studies.

### Professional competences to be acquired

### A Physicist:

### a) Knowledge:

- Is familiar with the theoretical and practical knowledge of the major topics of physics in the context and in the system level.

- Knowledge in field, laboratory and technical materials, instruments and methods of physics for high level practical work.

- Being familiar with high-level methods of scientific research, self-education and communication.

- Acquainted with the limitations and possible research directions in modern physics.

- Having a high level of knowledge of natural sciences and the elements of the practice that builds on it, and can systematize them.

- Knows the field, the laboratory, and the practical tools and methods related to physics, that enables him/her to perform a professional level work.

- Has a deep and thorough professional knowledge, which needs to solve problems related to natural processes, to natural resources, and to live and inanimate systems.

- Knows in its context the scientific problems, and the projects and systems, that can be treated.

- Familiar with the concepts and the terminology of modern physics and understands recent articles in the literature.

## b) Abilities:

- Is able to realize physical principles in natural phenomena, to perform their experimental investigation according to scientific standards, and to obtain theoretical understanding.

- Is able to join research groups to perform both basic and applied physics research.

- Is able to develop and operate industrial, IT, and measuring systems related to physical laws and high technology processes.

- Is able to use information technology in physics.

- With regular professional self-education, he/she is able to elaborate the new scientific results of physics, and apply them creatively in his/her research.

- Is able to test the projects and systems, that can be treated in his/her research field, with such methods that are accepted in phyics.

- Ability for planning, implementation and evaluation of experiments based on comprehensive knowledge attained in physics.

- Is able to ask questions about physics and related topics.

- Is able to continuously broaden her/his knowledge and has the aptitude for continuing her/his studies in the framework of doctoral studies.

- His/her knowledge and problem solving skills acquired during his/her education make him/her capable of doing independent and leader works in further fields, that use the scientific results and methods of physics (administration, environmental protection, etc.).

## c) Attitude:

- Basic gained properties are creativity, flexibility, problems recognizing and solving ability, intuition, systematic data processing ability.

- Seeks to familiarize herself/himself with new developments of modern physics and pursues the application of those developments in broad contexts.

- Able to distinguish well-established science from statements unsupported scientifically.

- He/she is characterized by the sensitivity to the environment, the positive attitude for the vocational training, and commitment for the quality work.

- Has ability for initiation, decision-making, and personal responsibility.

- Actively collaborates with his/her colleagues, contributes to teamwork in a constructive way, is able to assume leadership duties with sufficient experience.

- Able to explain the open questions of his field to experts as well as laymen.

- Constantly striving to expand his/her knowledge and acquire new skills.

## d) Autonomy and responsibility:

- Shows a large degree of independence in formulating his/her professional views on both broad and specialized topics of modern physics, as well as in the justification and representation of his/her professional views.

- Awares and takes responsibility for the scientific worldview.

- Based on his/her high level knowledge of physics, and on his/her critical and system level way of thinking, he/she can cooperate with the representatives of his/her own or other scientific areas in a responsible way.

- He/she is environmentally conscious during field and laboratory activities.

- Performs scientific research with the highest level of ethical standards.
- Forms his/her opinion awaring of the importance of the scientific thinking and conceptualization.

## **Completion of the MSc Program**

The Credit System

Majors in the Hungarian Education System have generally been instituted and ruled by the Act of Parliament under the Higher Education Act. The higher education system meets the qualifications of the Bologna Process that defines the qualifications in terms of learning outcomes: statements of what students know and can do on completing their degrees. In describing the cycles, the framework uses the European Credit Transfer and Accumulation System (ECTS).

ECTS was developed as an instrument of improving academic recognition throughout the European Universities by means of effective and general mechanisms. ECTS serves as a model of academic recognition, as it provides greater transparency of study programs and student achievement. ECTS in no way regulates the content, structure and/or equivalence of study programs.

Regarding each major the Higher Education Act prescribes which professional fields define a certain training program. It contains the proportion of the subject groups: natural sciences, economics and humanities, subject-related subjects and differentiated field-specific subjects.

During the program students have to complete a total amount of 120 credit points. It means approximately 30 credits per semester. The curriculum contains the list of subjects (with credit points) and the recommended order of completing subjects which takes into account the prerequisite(s) of each subject. You can find the recommended list of subjects/semesters in chapter "Model Curriculum of Environmental Science MSc Program".

### Requirements

The unit of measurement of academic requirements is the academic point (credit). Credits are used for evaluating the performance of education duties of students in programme. Students shall complete a total of at least 120 credits during the entire programme.

Precondition for taking the state exam for MSc students is to prepare the thesis which is the creative elaboration of a research task in written form as defined in the requirements of the training program. Students will finish their studies in the Physicist MSc program by taking the state exam which consists of two parts. On the one hand students have to present the most important elements of their thesis and have to defend it in the form of answering the questions of the state exam board. On the other hand, they have to take the oral exam on the same day.

### **Professional features**

Students have to take 120 credits:

- 4-16 credits of general knowledge from the fields of mathematics, informatics;

- 20-30 credits of general knowledge in the fields of modern physics (6 credits maximum from atomic and molecular physics, condensed matter physics, statistical physics, 9 credits maximum form nuclear and particle physics, 8 credits maximum from advanced laboratory courses);

- 30-60 credits of specialized knowledge in modern physics;

- 30 credits of thesis;

- 6 credits of optional subjects of the faculty;
- the compulsory vocational courses are divided into four blocks of 15-15 credits each.

#### Work and Fire Safety Course

According to the Rules and Regulations of University of Debrecen a student has to complete the online course for work and fire safety. Registration for the course and completion are necessary for graduation.

Registration in the Neptun system by the subject: MUNKAVEDELEM

Students have to read an online material until the end to get the signature on Neptun for the completion of the course. The link of the online course is available on webpage of the Faculty.

#### **Physical Education**

According to the Rules and Regulations of University of Debrecen a student has to complete Physical Education courses at least in one semesters during his/her Master's training. Our University offers a wide range of facilities to complete them. Further information is available from the Sport Centre of the University, its website: <u>http://sportsci.unideb.hu</u>. The physical education course is 1 credit/semester.

You can register for a PE course in Neptun within the deadline.

Exemptions may be requested on the basis of health grounds or proven competitive sporting activity. Requests for exemption must be submitted using the form available at www.sport.unideb.hu.

The deadlines are: 30 September and 28 February.

Location: University of Debrecen Institute for Sports Science Coordination

#### **Occupational safety**

Completion of the course in Occupational Health is a prerequisite for the final certificate (diploma). The course is 1 credit/semester.

#### **Pre-degree Certification**

A pre-degree certificate is issued by the Faculty after completion of the master's (MSc) program. The pre-degree certificate can be issued if the student has successfully completed the study and exam requirements as set out in the curriculum, the requirements relating to Physical Education as set out in Section 10 in Rules and Regulations, internship (mandatory) – with the exception of preparing thesis – and gained the necessary credit points (120). The pre-degree certificate verifies (without any mention of assessment or grades) that the student has fulfilled all the necessary study and exam requirements defined in the curriculum and the requirements for Physical Education. Students who obtained the pre-degree certificate can submit the thesis and take the final exam.

### Final exam

#### **Objectives**:

The professional competencies and knowledge of the student is evaluated. The graduate needs to demonstrate proficiency in physics and the ability to perform high level problems in physics independently. The preparedness for professional debates also needs to be shown.

### **Requirements**:

In order to participate in the final exam, the student needs to satisfy all formal and informal requirements. Thus, a minimum of 120 credits need to be earned. Further requirement is the submission of a thesis covering the diploma work of the graduate well before the final exam.

#### The diploma work

It is based on the independent research activity of the student under the guidance of the supervisor. Students have to choose a topic for their diploma work in the 2nd semester. They have to write it in two semesters. The thesis should be more than 25 pages. The cover page has to contain the name of the institute, the title of the thesis, the name and the degree program of the student, the name and the university rank of the supervisor. Besides the detailed discussion of the topic, the thesis should contain an introduction, a table of contents and a bibliography. The thesis has to be defended in the final exam.

#### The final exam

The final exam has two parts. First, the thesis based on the diploma work is presented in front of the examining committee. The graduating student gives a lecture, answers the remarks of the reviewer and the questions of the committee and the audience. In the second part, the student must demonstrate her/his knowledge in physics at the masters level. There are the following groups of questions:

- A: fundamental topics
- B: module topics

### Final Exam Board

Board chair and its members are selected from the acknowledged internal and external experts of the professional field. Traditionally, it is the chair and in case of his/her absence or indisposition the vice-chair who will be called upon, as well. The board consists of – besides the chair – at least two members (one of them is an external expert), and questioners as required. The mandate of a Final Examination Board lasts for one year.

### Repeating a failed Final Exam

If any part of the final exam is failed it can be repeated according to the rules and regulations. A final exam can be retaken in the forthcoming final exam period. If the Board qualified the Thesis unsatisfactory a student cannot take the final exam and he has to make a new thesis. A repeated final exam can be taken twice on each subject.

#### Diploma

The diploma is an official document decorated with the coat of arms of Hungary which verifies the successful completion of studies in the Physicist Master Program. It contains the following data: name of HEI (higher education institution); institutional identification number; serial number of diploma; name of diploma holder; date and place of his/her birth; level of qualification; training program; specialization; mode of attendance; place, day, month and year issued. Furthermore, it has to contain the rector's (or vice-rector's) original signature and the seal of HEI. The University keeps a record of the diplomas issued.

In the Physicist Master Program the diploma grade is calculated as the average grade of the results of the followings:

- Weighted average of the overall studies at the program (A)
- Average of grades of the thesis and its defense given by the Final Exam Board (B)
- Average of the grades received at the Final Exam for the two subjects (C)

Diploma grade = (A + B + C)/3

Classification of the award on the bases of the calculated average:

Excellent	4.81 - 5.00
Very good	4.51 - 4.80
Good	3.51 - 4.50
Satisfactory	2.51 - 3.50
Pass	2.00 - 2.50

# Structure of the curriculum

Knowledge elements,	semesters	semesters					evaluation
lectures and lecturers	1.	2.	credit				
	contact ho – seminar,	contact hours, types of teaching (1 – lecture, s – seminar, p – practice, c – consultation, lab –				points	
	laboratory	<i>(</i> )					

# Basic knowledge elements

Dr. András Csehi

Quantum mechanics – Statistical physics – Particle physics knowledge elements						
1. Quantum mechanics 2.	281,				4	exam
Dr. Sándor Nagy	14p/4cr					
2. Statistical physics 2.	281,				4	exam
Dr. Ferenc Kun	14p/4cr					
3. Particle physics 1.	281,				4	exam
Dr. Zoltán Trócsányi	14p/4cr					
<b>Environmental Physics</b>	– Atomic a	nd molecul	ar physics –	Condensed	matter kno	wledge elements
1. Laboratory of	42lab/4cr				4	mid-semester
environmental physics						grade
Dr. Zoltán Papp						
2. Atomic and molecular	281,				4	exam
physics I	14p/4cr					
Dr. András Csehi						
3. Condensed matter 3.	281,				4	exam
Dr. Erdélyi Zoltán	14p/4cr					

Knowledge elements of the	modules			
Fundamental interactions (	(Dr. Ádám Kardos)			
1. Theory of relativity	281,		5	exam
Dr. Zsolt Schram	14p/5cr			
2. Quantum field theory	281,		6	exam
Dr. Ádám Kardos	14p/6cr			
3. Particle physics 2.		281,	4	exam
Dr. Zoltán Trócsányi		14p/4cr		
4. Standard model		281,	4	exam
Dr. Zoltán Trócsányi		14p/4cr		
Atomic and molecular phy	sics and és quantum	informatics (I	Dr. Ágnes Vibók)	
1. Electronic structure	281,		5	exam
methods and quantum	14p/5cr			
dynamics				
Dr. Ágnes Vibók				
2. Atoms and molecules		281,	5	exam
in electromagnetic field		14p/5cr		
Dr. Ágnes Vibók		_		
3. Atomic and molecular	281,		5	exam
physics II	14p/5cr			
	1	1	1	1

1. Physics of complex systems       281, 14p/5cr       5       exam         Dr. Ferenc Kun       14p/5cr       -       -         2. Computer modelling       281, 14p/5cr       5       exam         Dr. Ferenc Kun       14p/5cr       -       -         Dr. Zsolt Gulácsi       -       -       -         4. Complex networks and applications       281, 14p/5cr       5       exam         Dr. Sahot Oxagy       -       -       -         Dr. Sahot Oxagy       -       -       -         Dr. Sahot Oxagy       -       -       -         Dr. Sahot Oxagy       -       -       -       -         Dr. Sahot Oxagy       -       -       -       -       -         Dr. Sahot Oxagy       -       -       -       -       -       -         Dr. Sahot Oxagy       -       <	Complex systems and statistical physics (Dr. Ferenc Kun)						
systems 12. Computer modelling 281, 14p/5cr 12. Computer modelling 281, 5 exam 14p/5cr 12. Computer modelling 281, 5 exam 14p/5cr 12. Computer modelling 281, 5 exam 14p/5cr 12. Complex networks and applications 14p/5cr 12. Complex networks and applications 14p/5cr 12. Condensed matter (Dr. Zoltán Erdélyi) 1. Nanodiffusion and segregation Dr. Zoltán Erdélyi 28ea/3cr 3eeaam 28eaam 28eaam 28ea/3cr 28ea/3cr 3eeaam 28eaam 28eaam 28ea/3cr 3eeaam 28eaam 28eaam 28eaam 28eaam 28eaam 28ea/3cr 3eeaam 28eaam 28eaam 28eaam 28eaam 28eaam 28ea/3cr 3eeaam 28eaam 28eaam 28eaam 28eaam 28ea/3cr 3eeaam 28eaam 28eaam 28eaam 28ea/3cr 3eeaam 28eaam 28e	1. Physics of complex			281,		5	exam
Dr. Ferenc Kun 281, 5 exam 281, 14p/Scr 281, 5 exam 77, 281, 14p/Scr 27, 281, 14p/Scr 281, 14p, 5 exam 14p/Scr 190,	systems			14p/5cr			
2. Computer modelling 281, 14p/5cr 5 exam Dr. Ferenc Kun 14p/5cr 5 exam critical phenomena 1. 14p/5cr 5 exam d. Complex networks and 281, 5 exam applications 14p/5cr 5 exam Dr. Zsolf Gulácsi 4 281, 5 exam applications 14p/5cr 7 8 exam Dr. Sándor Nagy 7 exam Condensed matter (Dr. Zoltán Erdélyi) 1. Nanodiffusion and 28ea/3cr 8 exam segregation 2 exam Dr. Zoltán Erdélyi 4 21ab/5cr 8 exam anomagnetism And 141, 5 mid-semester grade 281,14p, 5 exam anomagnetism KV 14 hab/5cr 9 exam anomagnetism KV 14 hab/5cr 9 exam anomagnetism KV 14 hab/5cr 9 exam ananomagnetism 8 144,25cr 9 exam ananytical 141,5cr 9 exam ananytical 144,5cr 9 exam Dr. Caltán Erdélyi 9 exam 14 hab/5cr 9 exam ananytical 144,5cr 9 exam 14 hab/5cr 9 exam 15 hab/	Dr. Ferenc Kun						
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Dr. Zsolt Gulácsi	critical phenomena 1.		14p/5cr				
4. Complex networks and applications       281, 14p/5cr       5       exam         pr. Sándor Nagy       Condensed matter (Dr. Zoltán Erdélyi)       3       exam         1. Nanodiffusion and segregation       28ea/3cr       3       exam         Dr. Zoltán Erdélyi       28ea/3cr       3       exam         2. Computer simulation       141, 42lab/5cr       5       mid-semester         Dr. Zoltán Erdélyi       28la/3cr       28l,14p, 5       exam         3. Magnetism and nanomagnetism       14       ab/5cr       mid-semester         T. Lajos Daróczi       14       ab/5cr       mid-semester         T. Csaba Cserháti       28lab/2cr       2       mid-semester         S. Transmission and analytical electronmicroscopy       28l,14p, 5       exam       exam         Dr. Erdélyiné Dr. Eszter       28l, 14p/4cr       4       exam         3. Radiation of       28l, 14p/4cr       4       exam         3. Radiation Protection       28l, 14p/4cr       4       exam         Dr. Erdélyiné Dr. Eszter       14p/4cr       4       exam         3. Radiation Protection       28l, 14p/4cr       4       exam         Dr. Erdélyiné Dr. Eszter       14lab/4cr       4       exam	Dr. Zsolt Gulácsi						
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KV       lab/5cr       lab/5cr         Dr. Lajos Daróczi       2       mid-semester         4. Measurement of       28lab/2cr       2       mid-semester         materials properties       28lab/2cr       2       mid-semester         Dr. Csaba Cserháti       2       14lab/5cr       5       exam         S. Transmission and       28l,14p,       5       exam         analytical       14lab/5cr       5       exam         electronmicroscopy       Dr. Eszbar Cserháti       5       exam         T. Environmental Physics (Dr. István Csige)       28l,       4       exam         1. Environmental Physics       28l,       4       exam         3.       14p/4cr       8       8         2. Simulation of       28l,       4       exam         environmental processes       14p/4cr       8       8         Dr. Erdélyiné Dr. Eszter       8       8       8       8         3. Radiation Protection       28l,       4       exam         and Dosimetry       14lab/4cr       9       9       9         4. Physics of the       28ea/3cr       3       8       8         5. Measurements in       28lab/2cr	nanomagnetism				14		
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environmental processes Dr. Erdélyiné Dr. Eszter Baradács 3. Radiation Protection and Dosimetry Dr. Zoltán Papp 4. Physics of the atmosphere Dr. István Csige 5. Measurements in Environmental Physics Dr. Zoltán Papp	2. Simulation of			281,		4	exam
Dr. Erdélyiné Dr. Eszter Baradács281, 281, 14lab/4cr4exam3. Radiation Protection and Dosimetry Dr. Zoltán Papp281, 14lab/4cr4exam4. Physics of the atmosphere Dr. István Csige28ea/3cr3exam5. Measurements in Environmental Physics Dr. Zoltán Papp28lab/2cr2mid-semester grade	environmental processes			14p/4cr			
BaradácsImage: Second StructureBaradácsImage: Second Structure3. Radiation Protection and Dosimetry281, 14lab/4cr4examDr. Zoltán Papp14lab/4crImage: Second Structure44. Physics of the atmosphere28ea/3cr3examDr. István Csige28lab/2cr2mid-semester grade5. Measurements in Environmental Physics28lab/2cr2mid-semester grade	Dr. Erdélyiné Dr. Eszter			1			
3. Radiation Protection and Dosimetry281, 14lab/4cr4examDr. Zoltán Papp14lab/4cr4exam4. Physics of the atmosphere28ea/3cr3examDr. István Csige28lab/2cr2mid-semester grade5. Measurements in Environmental Physics28lab/2cr2mid-semester grade	Baradács						
and Dosimetry Dr. Zoltán Papp14lab/4cr14lab/4cr4. Physics of the atmosphere Dr. István Csige28ea/3cr3exam5. Measurements in Environmental Physics Dr. Zoltán Papp28lab/2cr2mid-semester grade	3. Radiation Protection		281,			4	exam
Dr. Zoltán Papp28ea/3cr3exam4. Physics of the atmosphere28ea/3cr3examDr. István Csige28lab/2cr2mid-semester grade5. Measurements in Environmental Physics28lab/2cr2mid-semester grade	and Dosimetry		14lab/4cr				
4. Physics of the atmosphere28ea/3cr3examDr. István Csige28lab/2cr2mid-semester grade5. Measurements in Environmental Physics28lab/2cr2mid-semester grade	Dr. Zoltán Papp						
atmosphereImage: Constraint of the second secon	4. Physics of the		28ea/3cr			3	exam
Dr. István Csige28lab/2cr2mid-semester5. Measurements in Environmental Physics28lab/2cr2gradeDr. Zoltán Papp22grade	atmosphere		-				
5. Measurements in Environmental Physics28lab/2cr2mid-semester gradeDr. Zoltán Papp21111	Dr. István Csige						
Environmental Physics Dr. Zoltán Papp	5. Measurements in		28lab/2cr			2	mid-semester
Dr. Zoltán Papp	Environmental Physics						grade
	Dr. Zoltán Papp						

6. Nuclear analytical			28ea/3cr		3	exam
methods in environmental						
research						
Dr. Mihály Molnár						
Nuclear physics (Dr. Gy	rula Zilizi)					
1. Advanced Nuclear			281,		4	exam
Physics			14p/4cr			
Dr. Judit Darai						
2. Nuclear Technology			281,		4	exam
Dr. László Oláh			14p/4cr			
3. Nuclear Astrophysics		28ea/3cr			3	exam
Dr. Zsolt Fülöp						
4. Nuclear physics		56lab/4cr			4	mid-semester
laboratory						grade
Dr. Gyula Zilizi						
5. Insight into the present			281,		4	exam
day nuclear research			14p/4cr			
Dr. Judit Darai						
Quantum mechanical ma	any-body s	ystems (Dr.	Zsolt Gulá	csi)		
1. Quasiparticles in solid		281, 14p,			5	exam
state phyiscs		14lab/5cr				
Dr. István Nándori						
2. Condensed matter		281,			5	exam
physics 4.		14p/5cr				
Dr. Zsolt Gulácsi						
3. Quantum mechanical			281,		5	exam
many-body physics I			14p/5cr			
Dr. Gulácsi Zsolt						
4. Basics of Functional				281,	5	exam
Renormalization Group				14p/5cr		
Method						
Dr. István Nándori						
Diploma work						mid competer

Diploma work					mid-semester
		15p/10 cr	30p/20cr		grade
				30	

Title of course: Quantum Mechanics 2         Code: TTFME0102	ECTS Credit points: 4
Type of teaching, contact hours - lecture: 2 hours/week - practice: 1 hours/week - laboratory: -	
Evaluation: exam	
Workload (estimated), divided into contact hours: - lecture: 28 hours - practice: 14 hours - laboratory: - - home assignment: 40 hours - preparation for the exam: 40 hours Total: 122 hours	
Year, semester: 1 <sup>st</sup> year, 1 <sup>st</sup> semester	
Its prerequisite(s): -     Further courses built on it: -	
Topics of course	
Addition of angular momenta. Quantum statistical me approximation. Time-independent perturbation theory, non- Variational methods. Time-dependent perturbation theory. T Born approximation, optical theorem. Resonance scattering. D identical particles. Propagators, transition amplitude. Path Relativistic quantum mechanics, Dirac equation, Lorentz syn equation for free particles.	echanics. Classical limit, WKB degenerate and degenerate cases. he Lippmann-Schwinger equation, biscrete and continuous symmetries, integral in quantum mechanics. mmetry. The solution of the Dirac
Literature	
<i>Compulsory:</i> J. J. Sakurai, Modern Quantum Mechanics (Addison-Wesley, <i>Recommended:</i> James D. Björken, Sidney D. Drell, Relativistic Quantum Mec	2011) chanics (McGraw-Hill, 1964)
Schedule:	
Addition of angular momenta, the Clebsch-Gordan coefficient $2^{nd}$ week	s.
Quantum statistical mechanics, canonical and grand canonical $3^{rd}$ week Interpretations of the wave function, classical limit, semiclassi	ensembles. cal (WKB) approximation.
Time-independent perturbation theory, nondegenerate case. W $5^{th}$ week	Vave function renormalization.

Time-independent perturbation theory, degenerate case. Variational methods.

 $6^{th}$  week

Time-dependent perturbation theory, interaction picture, Dyson series. Transition probability, Fermi's golden rule.

 $7^{th}$  week

The Lippmann-Schwinger equation, Born approximation, optical theorem.

 $8^{th}$  week

Method of partial waves. Resonance scattering.

9<sup>th</sup> week

Discrete symmetries: parity, time-reversal symmetry. Continuous symmetries, conservation laws.  $10^{th}$  week

Propagators, transition amplitude.

11<sup>th</sup> week

Feynman's path integrals in quantum mechanics.

 $12^{th}$  week

Potentials and gauge transformations, Aharonov-Bohm effect, gravity in quantum mechanics.

13<sup>th</sup> week

Relativistic quantum mechanics. Dirac equation, Lorentz symmetry.

14<sup>th</sup> week

The solution of the Dirac equation for free particles. Non-relativistic limit, Foldy-Wouthuysen transformation.

### **Requirements:**

- for a signature

Attendance at **lectures** is recommended, but not compulsory.

Participation at **practice classes** is compulsory. A student must attend the practice classes and may not miss more than three times during the semester. In case a student does so, the subject will not be signed and the student must repeat the course.

During the semester there are practical home works which have to be evaluated and submitted by the end of the  $14^{th}$  week of the semester. The requirement for a signature is a successful (> 50%) completion of the home works.

- for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have a practical signature.

The grade for the examination is given according to the following table:

Score	Grade
0-49	fail (1)
50-62	pass (2)
63-75	satisfactory (3)
76-88	good (4)
89-100	excellent (5)

### Person responsible for course: Dr. Sándor Nagy, associate professor, PhD

Lecturer: Dr. Sándor Nagy, associate professor, PhD

<b>Title of course</b> : Statistical physics 2 <b>Code</b> : TTFME0103	<b>ECTS Credit points:</b> 4
Type of teaching, contact hours	i
- lecture: 2 hours/week	
- practice: 1 hours/week	
- laboratory: -	
Evaluation: exam	
Workload (estimated), divided into contact hours:	
- lecture: 28 hours	
- practice: 14 hours	
- laboratory: -	
- home assignment: 40 hours	
- preparation for the exam: 40 hours	
Total: 122 hours	
<b>Year, semester</b> : 1 <sup>st</sup> year, 1 <sup>st</sup> semester	
Its prerequisite(s): -	
Further courses built on it: -	
Topics of course	
Fluctuations in thermal equilibrium. Time dependent fluc correlation of fluctuations. Spectral decomposition, the spe Correlations and response functions. The generalized susce fluctuations. Fluctuation-dissipation theorem. The theory of Linear transport, transport coefficients, entropy producted laws. Cross effects, Seebeck- and Peltier-effects, the Th hypothesis, microscopic reversibility, Onsager's reciprocit Brownian motion, random walk. Einstein-model. Smolu- laws. Conduction phenomena in terms of Brownian motion Diffusion processes, Fokker–Planck-equation. Langevin-e Fluctuation-dissipation theorem and its physical consequent Master equation, stationary solutions. Relation of the Master methods. Boltzmann transport equation, collision integral. Electric c	ctuations, correlation functions, spatial ectral density. Wiener-Hinchin theorem. eptibility and its relation to equilibrium of linear response. on. Electric and heat conduction, Fick homson relation. Onsager's regression ty relations. chowski description. Diffusion, Fick's n, Nyquist-noise. equation, Ornstein–Uhlenbeck-process. nces. ter equation to Monte Carlo simulations conduction in classical ideal gases.
Literature	
<i>Compulsory:</i> Sailer Kornél, Statisztikus fizika 2. (egyetemi jegyzet, Del R. Kubo, M. Toda, and N. Hashitsume, Statistical Ph Mechanics (Springer Verlag, Heidelberg, 1985). P.L. Krapivsky, S. Redner, and E. Ben-Naim, A Kinetic V University Press, 2010).	breceni Egyetem). 1ysics II – Nonequilibrium Statistical View of Statistical Physics (Cambridge
Schedule:	
1 <sup>st</sup> week	
Basics of equilibrium statistical physics. Ensembles, statis	stical averages. Fluctuation of the energy,

dependence of fluctuations. Equivalence of ensembles.

 $2^{nd}$  week

Spectral decomposition of fluctuations, power spectrum and the spectral density. The Wiener-Hinchin-theorem.

 $\mathcal{3}^{rd}$  week

Correlations and response functions. Generalized susceptibility. Fluctuation-dissipation theorem.  $4^{th}$  week

Theory of linear response. Linear transport, transport coefficients. Electric and heat conduction, Fick laws. Cross effects, Seebeck- and Peltier-effects, Thomson relation.

 $5^{th}$  week

Onsager's regression hypothesis, microscopic reversibility, Onsager's reciprocity law.

 $6^{th}$  week

Brownian motion, Einstein-model. Diffusion. Smoluchowski description. Conduction phenomena in terms of Brownian motion, Nyquist-noise.

 $7^{th}$  week

Generalized random walks, Levy flights. Anomalous diffusion.

 $8^{th}$  week

Diffusion processes, Fokker–Planck-equation, Wiener-process, Gaussian white noise.

9<sup>th</sup> week

Langevin-equation and its solutions, equivalent Fokker–Planck-equation. Ornstein–Uhlenbeck-process.  $10^{th}$  week

Fluctuation-dissipation theorem and its consequences. Applications of the Langevine equation.

 $11^{th}$  week

Master equation, stationary solutions. Relation of the Master equation to Monte Carlo simulations methods.

 $12^{th}$  week

Applications of the Master equation. Aggregation phenomena, gelation. Fragmentation processes. Growing complex networks. Time evolution and stationary solutions.

13<sup>th</sup> week

Boltzmann-transport equation, collisional integral. Electric conduction of the classical ideal gas. H-theorem and its relation to the entropy.

14<sup>th</sup> week

Applications of the Boltzmann-equation for simple problems: Dissipative gases of granular materials, inelastic collapse, driven dissipative gases. Ballistic agglomeration. Single lane traffic problems, emergence of traffic jams on highways.

### **Requirements:**

- for a signature

Attendance at **lectures** is recommended, but not compulsory.

Participation at **practice classes** is compulsory. A student must attend the practice classes and may not miss more than three times during the semester. In case a student does so, the subject will not be signed and the student must repeat the course.

During the semester there are practical home works which have to be evaluated and submitted by the end of the 14<sup>th</sup> week of the semester. The requirement for a signature is a successful completion of all the home works.

- for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have a practical signature.

The grade for the examination	is given according to the following table:		
Score	Grade		
0-49	fail (1)		
50-62	pass (2)		
63-75	satisfactory (3)		
76-88	good (4)		
89-100	excellent (5)		
Person responsible for course: Dr. Ferenc Kun, professor, PhD			
Lecturer: Dr. Ferenc Kun, professor, PhD			

Title of course: Particle Physics 1 Code: TTFME0104	ECTS Credit points: 4
Type of teaching, contact hours	•
- lecture: 2 hours/week	
- practice: 1 hours/week	
- laboratory: -	
Evaluation: exam	
Workload (estimated), divided into contact hours:	
- lecture: 28 hours	
- practice: 14 hours	
- laboratory: -	
- home assignment: 40 hours	
- preparation for the exam: 40 hours	
Total: 122 hours	
<b>Year, semester</b> : 1 <sup>st</sup> year, 1 <sup>st</sup> semester	
Its prerequisite(s): -	
Further courses built on it: -	
Topics of course	
Symmetries and particles. Static quark model. Free fermion. Experim model. Particle accelerators. Particle detectors. Event registration. violation, anomalous magnetic moment). CP, CPT and kaons. Neutr applications. Basics of cosmology.	ental verification of quark Basic experiments (parity rino experiments. Medical
Literature	
Compulsory: Dezső Horváth and Zoltán Trócsányi: Intorduction to Particle Phy Publishing, 2019. <i>Recommended:</i> Donald H. Perkins: Introduction to High Energy Physics, Addison-We Francis Halzen and Alan D.~Martin: Quarks and Leptons. An Introd Particle Physics, John Wiley and Sons, New York.	esley, Menlo Park, USA ductory Course in Modern
Schedule:	
1 WEEN	

Structure and details of the course, recommended literature and the available information on the web.

 $2^{nd}$  week

Fermions and bosons, leptons and hadrons. Interactions and medating bosons. Symmetries and conserved quantities, Noether's theorems. Particle reactions and Feynman diagrams. Spin, isospin and SU(2) symmetry. Strangeness, SU(3) and the first three quarks.

 $3^{rd}$  week

Coloured quarks. SU(3) algebra. Basics of quantum colour dynamics (QCD). Hadrons, mesons and baryons. Quark multiplets and excited states.

 $4^{th}$  week

Covariant formalism. Dirac equation and Lagrange functions. U(1) invariance, fermion current and continuity. Nucleon as quark atom. Magnetic moment of the neutron.

5<sup>th</sup> week

Cross section, resonance. Invariant mass. Width of the Z boson and the three fermion families. The fractional charge and colours of quarks. Overview of the quark model.

 $6^{th}$  week

Accelerator types. Storage rings, beam cooling. CERN's accelerator complex. Penetration of charged particles in matter.

 $7^{th}$  week

Multiwire chambers, scintillator and Cherenkov detectors. Transition radiation. Overview and comparison of the detector types. Structure of the CMS detector at the LHC.

 $8^{th}$  week

Particle identification. Detector calibration. Event registration and trigger logic. OPAL detector and characteristic event types at LEP. Collider Detector at Fermilab and its strange events. LHC: Observation of the Higgs boson.

9<sup>th</sup> week

Mirror symmetries. Discovery of parity violation. Muon spin resonance and its applications. Renormalisation of electromagnetic charge. Anomalous magnetic moment of the muon. Measuring  $(g - 2)_{\mu}$ : CERN  $\rightarrow$  BNL  $\rightarrow$  FNAL.

10<sup>th</sup> week

CP and CPT . Kaon regeneration, strangeness oscillation. Indirect and direct CP violation. Tests of CPT invariance.

11<sup>th</sup> week

Neutrino production. The Reines-Cowan experiment. Neutrino oscillation. Modern experiments: Superkamiokande, SNO, Ice Cube, Daya Bay. The LSND mystery. Long distance experiments. How many neutrinos? Neutrino problems and mysteries.

 $12^{th}$  week

Particle physics applications: basic research vs. applications. Medical diagnostics using particle accelerators. Isotope therapy and diagnosis. Accelerator therapy, hadron therapy.

13<sup>th</sup> week

Expanding universe, cosmologic principle. Big bang and inflation. Cosmic background radiation. Satellite observatories: COBE, WMAP, Planck and Hubble telescope. Dark matter and dark energy. Gravitational waves. Big bang and religion.

 $14^{th}$  week

Overview, discussion session.

#### **Requirements:**

- for a signature

Attendance at **lectures** is recommended, but not compulsory.

Participation at **practice classes** is compulsory. A student must attend the practice classes and may not miss more than three times during the semester. In case a student does so, the subject will not be signed and the student must repeat the course.

During the semester there are practical home works. The requirement for a signature is a successful (> 50%) completion of the home works.

- for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have a practical signature. The examination starts with a quiz: 8-12 questions with multiple choice answers. The students should be able to explain the chosen answers.

Scoring above 50%: satisfactory,

above 75%: average,

in addition, the explanations are correct as well: *good*, in addition, the interpretation of the knowledge is correct: *excellent*.

Person responsible for course: Dr. Zoltán Trócsányi, professor, PhD

Lecturer: Dr. Dezső Horváth, professor, PhD

Type of teaching, contact hours       - lecture: -         - practice: -       - jaboratory: 4 hours/week         Evaluation: grade for written laboratory record         Workload (estimated), divided into contact hours:         - lecture: -         - practice: -         - laboratory: 56 hours         - home assignment: 28 hours         - home assignment: 28 hours         - preparation for the exam: -         Total: 84 hours         Year, semester: 1 <sup>st</sup> year, 1 <sup>st</sup> semester         Is prerequisite(s): -         Further courses built on it: -         Topics of course         Determination of gamma-emitting radionuclides in soil (3x4 hours). Measurement of ambient alpha radioactivity by solid state nuclear track detector (2x4 hours). Measurement of tritum oncentration of environmental water samples by 3He method (3x4 hours). Madiocarbon dating and environmental research (2x4 hours). Sampling and analysis by PIXE method of atmospheric aerosols (2x4 hours). Measurement of stable isotope ratios by mass spectrometric method (2x4 hours). Neasurement of stable isotope ratios by mass spectrometric method (2x4 hours). Neat dating (2x4 hours). X-ray fluorescence analysis of environmental samples (2x4 hours). Measurement of the airborne concentrations of radon and radon progeny in different environments (2x4 hours). X-ray fluorescence analysis of environmental samples (2x4 hours). Determination of the airborne concentrations of radon and radon progeny in different environments (2x4 hours). X-ray fluorescence analysis of environmental samples (2x4 hours). Determination of the airborne concentr	<b>Title of course</b> : Laboratory of Environmental Physics <b>Code</b> : TTFML0106	ECTS Credit points: 4
<ul> <li>lecture: -         <ul> <li>practice: -                 <ul></ul></li></ul></li></ul>	Type of teaching, contact hours	
<ul> <li>- practice: -         <ul> <li>- laboratory: 4 hours/week</li> </ul> </li> <li>Evaluation: grade for written laboratory record</li> <li>Workload (estimated), divided into contact hours:         <ul> <li>- lecture: -</li> <li>- practice: -</li> <li>- laboratory: 56 hours</li> <li>- home assignment: 28 hours</li> <li>- preparation for the exam: -</li> <li>Total: 84 hours</li> </ul> </li> <li>Year, semester: 1<sup>st</sup> year, 1<sup>st</sup> semester</li> <li>Its prerequisite(s): -         <ul> <li>Further courses built on it: -</li> </ul> </li> <li>Determination of gamma-emitting radionuclides in soil (3x4 hours). Measurement of ambient alpha radioactivity by solid state nuclear track detector (2x4 hours). Measurement of tritium concentration of environmental water samples by 3He method (3x4 hours). Radiocarbon dating and environmental research (2x4 hours). Sampling and analysis by PIXE method of atmospheric aerosols (2x4 hours). Measurement of stable isotope ratios by mass spectrometric method (2x4 hours). K-Ar dating (2x4 hours). X-ray fluorescence analysis of environmental samples (2x4 hours). Determination of the airborne concentrations of radon and radon progeny in different environments (2x4 hours). X-ray fluorescence analysis of environmental samples (2x4 hours). Determination of the airborne concentrations of radon and radon progeny in different environments (2x4 hours). X-ray fluorescence analysis of environmental samples (2x4 hours). The content of the course in the given academic year is determined and announced by the responsible instructor based on the above offering, after negotiations with the practical tutors on the course with the dubic for the</li></ul>	- lecture: -	
<ul> <li>laboratory: 4 hours/week</li> <li>Evaluation: grade for written laboratory record</li> <li>Workload (estimated), divided into contact hours:         <ul> <li>lecture: -</li> <li>practice: -</li> <li>laboratory: 56 hours</li> <li>home assignment: 28 hours</li> <li>preparation for the exam: -</li> </ul> </li> <li>Total: 84 hours</li> <li>Year, semester: 1<sup>st</sup> year, 1<sup>st</sup> semester</li> <li>Its prerequisite(s): -</li> <li>Further courses built on it: -</li> <li>Topics of course</li> <li>Determination of gamma-emitting radionuclides in soil (3x4 hours). Measurement of ambient alpha radioactivity by solid state nuclear track detector (2x4 hours). Measurement of tritium concentration of environmental water samples by 3He method (3x4 hours). Radiocarbon dating and environmental research (2x4 hours). Sampling and analysis by PIXE method of atmospheric aerosols (2x4 hours). Measurement of stable isotope ratios by mass spectrometric method (2x4 hours). K-Ar dating (2x4 hours). X-ray fluorescence analysis of environmental samples (2x4 hours). Determination of the airborne concentrations of radon and radon progeny in different environments (2x4 hours).</li> <li>The content of the course in the given academic year is determined and announced by the responsible instructor based on the above offering, after negotiations with the practical tutors on the deve offering, after negotiations with the practical tutors on the deve offering.</li> </ul>	- practice: -	
Evaluation: grade for written laboratory record  Workload (estimated), divided into contact hours:  - lecture: practice: laboratory: 56 hours - home assignment: 28 hours - preparation for the exam: - Total: 84 hours  Year, semester: 1 <sup>st</sup> year, 1 <sup>st</sup> semester Its prerequisite(s): -  Further courses built on it: -  Determination of gamma-emitting radionuclides in soil (3x4 hours). Measurement of ambient alpha radioactivity by solid state nuclear track detector (2x4 hours). Radiocarbon dating and environmental research (2x4 hours). Sampling and analysis by PIXE method of atmospheric aerosols (2x4 hours). Measurement of stable isotope ratios by mass spectrometric method (2x4 hours). K-Ar dating (2x4 hours). X-ray fluorescence analysis of environmental samples (2x4 hours). K-Ar dating (2x4 hours). X-ray fluorescence analysis of environmental samples (2x4 hours). K-Ar dating (2x4 hours). X-ray fluorescence analysis of environmental samples (2x4 hours). Determination of the airborne concentrations of radon and radon progeny in different environments (2x4 hours). The content of the course in the given academic year is determined and announced by the responsible instructor based on the above offering, after negotiations with the practical tutors on the content versibilitie of the there the uber t	- laboratory: 4 hours/week	
Workload (estimated), divided into contact hours:         - lecture: -         - practice: -         - laboratory: 56 hours         - home assignment: 28 hours         - preparation for the exam: -         Total: 84 hours         Year, semester: 1st year, 1st semester         Its prerequisite(s): -         Further courses built on it: -         Topics of course         Determination of gamma-emitting radionuclides in soil (3x4 hours). Measurement of ambient alpha radioactivity by solid state nuclear track detector (2x4 hours). Measurement of tritium concentration of environmental water samples by 3He method (3x4 hours). Radiocarbon dating and environmental research (2x4 hours). Sampling and analysis by PIXE method of atmospheric aerosols (2x4 hours). Measurement of stable isotope ratios by mass spectrometric method (2x4 hours). K-Ar dating (2x4 hours). X-ray fluorescence analysis of environmental samples (2x4 hours). K-Ar dating (2x4 hours). X-ray fluorescence analysis of environmental samples (2x4 hours). Determination of the airborne concentrations of radon and radon progeny in different environments (2x4 hours).         The content of the course in the given academic year is determined and announced by the responsible instructor based on the above offering, after negotiations with the practical tutors on the determined home of the course in the given academic year is determined and announced by the responsible instructor based on the above offering, after negotiations with the practical tutors on the determined home of the course in the given academic year is determined and announced by the responsible instructor based on the above offering, after negotiations with the pr	Evaluation: grade for written laboratory record	
<ul> <li>lecture: - practice: - practice: - laboratory: 56 hours home assignment: 28 hours preparation for the exam: - Total: 84 hours Year, semester: 1<sup>st</sup> year, 1<sup>st</sup> semester Year, semester: 1<sup>st</sup> year, 1<sup>st</sup> semester Its prerequisite(s): - Further courses built on it: - Topics of course Determination of gamma-emitting radionuclides in soil (3x4 hours). Measurement of ambient alpha radioactivity by solid state nuclear track detector (2x4 hours). Radiocarbon dating and environmental research (2x4 hours). Sampling and analysis by PIXE method of atmospheric aerosols (2x4 hours). Measurement of stable isotope ratios by mass spectrometric method (2x4 hours). K-Ar dating (2x4 hours). X-ray fluorescence analysis of environmental samples (2x4 hours). K-Ar dating (2x4 hours). X-ray fluorescence analysis of environmental samples (2x4 hours). K-Ar dating (2x4 hours). X-ray fluorescence analysis of environmental samples (2x4 hours). Determination of the airborne concentrations of radon and radon progeny in different environments (2x4 hours).</li> <li>The content of the course in the given academic year is determined and announced by the responsible instructor based on the above offering, after negotiations with the practical tutors on </li> </ul>	Workload (estimated), divided into contact hours:	
<ul> <li>- practice: - <ul> <li>laboratory: 56 hours</li> <li>home assignment: 28 hours</li> <li>- preparation for the exam: -</li> </ul> </li> <li>Total: 84 hours</li> </ul> <li>Year, semester: 1st year, 1st semester Its prerequisite(s): - Further courses built on it: - Topics of course Determination of gamma-emitting radionuclides in soil (3x4 hours). Measurement of ambient alpha radioactivity by solid state nuclear track detector (2x4 hours). Measurement of tritium concentration of environmental water samples by 3He method (3x4 hours). Radiocarbon dating and environmental research (2x4 hours). Sampling and analysis by PIXE method of atmospheric aerosols (2x4 hours). Measurement of stable isotope ratios by mass spectrometric method (2x4 hours). K-Ar dating (2x4 hours). X-ray fluorescence analysis of environmental samples (2x4 hours). Determination of the airborne concentrations of radon and radon progeny in different environments (2x4 hours). The content of the course in the given academic year is determined and announced by the responsible instructor based on the above offering, after negotiations with the practical tutors on the same of the responsible instructor based on the above offering, after negotiations with the practical tutors on the same of the responsible instructor based on the above offering.</li>	- lecture: -	
<ul> <li>- laboratory: 56 hours</li> <li>- home assignment: 28 hours</li> <li>- preparation for the exam: -</li> <li>Total: 84 hours</li> <li>Year, semester: 1<sup>st</sup> year, 1<sup>st</sup> semester</li> <li>Its prerequisite(s): -</li> <li>Further courses built on it: -</li> <li>Topics of course</li> <li>Determination of gamma-emitting radionuclides in soil (3x4 hours). Measurement of ambient alpha radioactivity by solid state nuclear track detector (2x4 hours). Measurement of tritium concentration of environmental water samples by 3He method (3x4 hours). Radiocarbon dating and environmental research (2x4 hours). Sampling and analysis by PIXE method of atmospheric aerosols (2x4 hours). Measurement of stable isotope ratios by mass spectrometric method (2x4 hours). K-Art dating (2x4 hours). X-ray fluorescence analysis of environmental samples (2x4 hours). Determination of the airborne concentrations of radon and radon progeny in different environments (2x4 hours).</li> <li>The content of the course in the given academic year is determined and announced by the responsible instructor based on the above offering, after negotiations with the practical tutors on</li> </ul>	- practice: -	
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#### Literature

Compulsory:

Mandatory data sources for each practice required by the tutors in the given academic year. *Recommended:* 

For each practice, the data sources recommended by the tutors in the given academic year.

#### Schedule:

The themes cannot be recorded weekly in advance for each student because the individual exercises are done by students in pairs, the same week the different student pairs perform different exercises and the order of the exercises will be different for different student pairs. At the beginning of the term of the semester, a schedule is drawn up that determines which week the exercises are performed for each student pairs.

#### 1<sup>st</sup> exercise

Determination of gamma-emitting radionuclides in soil (3x4 hours). During the course, students will learn about the operation and use of semiconductor detector gamma spectrometers by determining the

content of gamma-emitting radionuclides in environmental samples, mainly in soil. For the first occupation, students should bring 150-200 g of raw soil samples per person, which comes from their living environment. During the first two sessions, students prepare their soil samples for measurement (drying, peeling, sifting) and, on the other hand, they gain insight into the physical background of the operation of the spectrometer determining the full energy peak efficiency vs. gamma energy curve for the geometry used to measure soil samples. On the third session they evaluate the gamma spectra taken for the soil samples and interpret the results obtained.

## 2<sup>nd</sup> exercise

Measurement of environmental alpha radioactivity by means of a solid state nuclear track detector (2x4 hours). The student will be informed of the method used to determine the activity concentration of the alpha-decaying radon and thoron gases in the environment by using solid state nuclear track detector. In the first session, the student produces 3 pairs of Radamon type measuring instruments, which include cutting off the track detector disks and assembling the measuring devices. Then installs the gauges to measure the radon and thoron content of the soil gas in situ. In the second session, the collected detectors are processed by etching, tracks are counted using an optical microscope, and the densities and average Rn and Tn activity concentrations are determined.

3<sup>rd</sup> exercise

Measurement of the tritium concentration of environmental water samples by using 3He method (3x4 hours). The amount to be measured is now between 0 and 20 TU. The detection limit of conventional liquid scintillation measurement technology (currently 4 TU) can be improved by electrolytic enrichment but does not reach 0.4 TU. The method used in the practice determines the amount of tritium not by its radioactive decay, but by the production of the daughter of 3H, the 3He. The water sample is first distilled off, so water is purified from most pollutants and dissolved salts. The water samples are then filled into stainless steel tanks and then vacuumed with high vacuum pumps to remove the air in the container and the gases dissolved in the water. So the water will be completely helium-free. The containers are then sealed. The water sample is only consumed in a few weeks/months, during which time part of the tritium is decays and becomes helium-3. After storage, 3He from the tritium decay in the container is deposited in a special noble gas mass spectrometer and its quantity is determined. Based on the known water mass, the storage time and the volume of the measured 3He, the tritium concentration of water can be calculated. The exercise consists of 3 parts: 1. sampling, distilling, removal of gases; 2. mass spectrometer measurements; 3. evaluation, applications.

### 4<sup>th</sup> exercise

Radiocarbon dating and environmental research (2x4 hours).

#### 5<sup>th</sup> exercise

Sampling and analysis by PIXE method of atmospheric aerosol (2x4 hours). The student learns about the methods of sampling of atmospheric aerosol, concentration determination and nuclear analysis techniques, and gains experience in carrying out measurements based on particle accelerator. Tasks: (1) taking atmospheric aerosol samples with a personal or portable sampler; (2) measuring the mass of the aerosol sample collected onto the filters, calculating the atmospheric aerosol concentration; (3) determination of the element composition of the atmospheric aerosol sample by particle-induced X-ray emission analytical method. Evaluation of spectra by suitable software, computation of concentration, separation of anthropogenic and natural components.

#### 6<sup>th</sup> exercise

Measurement of stable isotope ratios by mass spectrometric method (2x4 hours).

#### 7<sup>th</sup> exercise

K-Ar dating (2x4 hours). Students will become familiar with the theoretical and practical back-grounds of K-Ar and Ar-Ar radiometric dating methods, the limitations and possibilities of the methods, the measureable samples and their properties. They look at the wet-chemical laboratory, where sample preparation and K-measurement are underway, and get acquainted with the construction and operation of noble gas mass spectrometers. They determine the age of a selected and prepared sample.

8<sup>th</sup> exercise

X-ray fluorescence analysis of environmental samples (2x4 hours). Students learn about the theoretical basics and the practical background of the x-ray fluorescence analytical method based on X-ray excitation and X-ray spectrometry by Si (Li) detector. As an example, some elemental concentrations are measured in metal alloys and other samples of environmental origin and conclusions are drawn from the results deduced.

9<sup>th</sup> exercise

Investigation of the airborne concentration of radon and radon decay products in the laboratory, indoors and outdoors, estimation of radon dose. Devices used: compact portable radon monitors; a filter and filter head for filtering aerosols from the air; gas volume meter; manometer; vacuum cleaner; desktop beta counter; desktop computer with suitable software. The student measures the radon gas activity concentration in the 2nd floor laboratory room with compact radon monitors, takes a sample of the aerosol content of the laboratory air by filtration, measures the beta radiation of the aerosol sample as a function of time under well-defined conditions and determines the equilibrium equivalent activity concentration of the radon decay products. From the results, he calculates the equilibrium factor and estimates the annual value of the radon dose. The student performs measurements on the ground floor corridor of the laboratory building and then outdoors near the building also. The equilibrium factor is determined at both sites and an estimate of the annual value of the radon dose is provided. The results obtained at different sites are compared and conclusions are drawn based on this comparison.

#### **Requirements:**

- for a signature

Participation at laboratory sessions is compulsory. A student must attend all the four sessions. In case a student doesn't so, the course will not be signed and the student must repeat it. Attendance at laboratory sessions will be recorded by the session leader. Being late is equivalent with an absence. Students are required to bring drawing instruments to each sessions. Active participation is evaluated by the teacher. If a student's behavior or conduct doesn't meet the requirements of active participation, the teacher may evaluate his/her participation as an absence.

#### - for a grade

The student will obtain grades for all the exercises one by one. The grades go from fail (1) to excellent (5) according to the following table:

Score	Grade
0-40	fail (1)
41-55	pass (2)
56-70	satisfactory (3)
71-85	good (4)
86-100	excellent (5)

The grade of the course will be the arithmetic mean of the grades obtained for each exercises rounded to the full, provided that the student has completed all the exercises with a grade better than fail (1). If the latter condition is not met then the grade of the course is fail (1) and the student must repeat the course in conformity with the EDUCATION AND EXAMINATION RULES AND REGULATIONS.

Person responsible for course: Dr. Zoltán Papp, associate professor, PhD

Responsible instructor: Dr. Zoltán Papp, associate professor, PhD

<b>Title of course</b> : Atomic and molecular physics I. <b>Code</b> : TTFME0101	ECTS Credit points: 4
Type of teaching, contact hours	
- lecture: 2 hours/week	
- practice: 1 hours/week	
- laboratory: -	
Evaluation: exam	
Workload (estimated), divided into contact hours:	
- lecture: 28 hours	
- practice: 14 hours	
- laboratory: -	
- home assignment: 40 hours	
- preparation for the exam: 40 hours	
Total: 122 hours	
Year, semester: 1 <sup>st</sup> year, 1 <sup>st</sup> semester	
Its prerequisite(s): Quantum Mechanics	
Further courses built on it: -	
Topics of course	
Electronic structure of one-electron atoms. Fine- and hyperfice corrections. Alkali metals, many-electron atoms and multiplicity Zeeman-effect. Time-dependent Schrödinger equation in the Molecular Schrödinger equation, Born-Oppenheimer approxime rovibrational and electronic transitions. Selection rules. Elect molecular ion, molecular orbitals. Aspect of polyatomic molecular	ne structure, relativistic and spin-orbit /. LS- and jj-couplings. Stark-effect and presence of periodic external fields. ation. Spectra of diatomics: rotational, ronic structure of diatomics, the H2+ les. Molecular symmetry.
Literature	
<ul> <li><i>Compulsory:</i></li> <li>B. H. Bransden, C. J. Joachain: Physics of atoms and molecules</li> <li>D. J. Griffiths: Introduction to Quantum Mechanics, Prentice-Har <i>Recommended:</i></li> <li>I. N. Levine: Quantum Chemistry, Prentice Hall, 5 edition</li> </ul>	, Longman Scientific & Technical all: New Jersey
Schedule:	
I <sup>st</sup> week	
Electronic structure of one-electron atoms. Energy levels, wave the Bohr-model. Rydberg atoms. $2^{nd}$ week	functions, quantum numbers, relation to

Fine- and hyperfine structure of one-electron atoms: relativistic and spin-orbit corrections; isotope effect and nuclear spin. Lamb-shift.

 $3^{rd}$  week

Electronic structure of alkali metals. Doublet structure of atomic sodium, multiplicity. Angular momentum and magnetic moment. The Landé factor.

4<sup>th</sup> week

Electronic structure of many-electron atoms. Central field approximation, LS- and jj coupling schemes. The He atom.

5<sup>th</sup> week

Stationary electric field as a perturbation: linear and quadratic Stark effect; dipole moment and polarizability.

 $6^{th}$  week

Stationary magnetic field as a perturbation: weak and strong field, normal and anomalous Zeeman effect.  $7^{th}$  week

Time-dependent Schrödinger equation of charged particles in external electromagnetic fields. Transition probabilities.

 $8^{th}$  week

Schrödinger equation of molecular systems: separability of the nuclear and electronic motions. The Born-Oppenheimer approximation and its validity.

9<sup>th</sup> week

Spectra of diatomics: rotational transitions, rovibrational spectral lines, electronic transitions. Selection rules. The impact of translation.

 $10^{th}$  week

Electronic structure of diatomics. The example of the H2+ molecule: atomic- and molecular orbitals, electron densities, formation of the chemical bond.

 $11^{th}$  week

Higher excited states of the H2+ molecule. Bonding and antibonding orbitals. Hibridization, properties of molecular orbitals. Aspect of polyatomic systems.

 $12^{th}$  week

Symmetry properties of molecules: symmetry elements, symmetry operations. Point groups. Symmetry of molecular orbitals.

13<sup>th</sup> week

Summary and consultation.

14<sup>th</sup> week

End-term test.

#### **Requirements:**

- for a signature

Attendance at **lectures** is recommended, but not compulsory.

Participation at **practice classes** is compulsory. A student must attend the practice classes and may not miss more than three times during the semester. In case a student does so, the subject will not be signed and the student must repeat the course.

During the semester there are practical home works which have to be evaluated and submitted by the end of the  $14^{th}$  week of the semester. The requirement for a signature is a successful (> 50%) completion of the home works.

- for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have a practical signature.

The grade for the examination is given according to the following table:

Score	Grade
0-49	fail (1)
50-62	pass (2)
63-75	satisfactory (3)
76-88	good (4)
89-100	excellent (5)

Person responsible for course: Dr. András Csehi, assistant professor, PhD

Lecturer: Dr. András Csehi, assistant professor, PhD

Title of course: Condensed matter III Code: TTFME0105	ECTS Credit points: 3	
Type of teaching, contact hours		
- lecture: 2 hours/week		
- practice: 1 hours/week		
- laboratory: -		
Evaluation: exam		
Workload (estimated), divided into contact hours:		
- lecture: 28 hours		
- practice: 14 hours		
- laboratory: -		
- home assignment: 40 hours		
- preparation for the exam: 40 hours		
Total: 122 hours		
Year, semester: 1 <sup>st</sup> year, 1 <sup>st</sup> semester		
Its prerequisite(s): -		
Further courses built on it: TTFME0141, TTFML0144		
Topics of course		
Phase Diagrams: Thermodynamic descriptions of binary systems, understanding the basic concept of the Calphad method; Diffusion under external driving forces: Nernst-Einstein equation, thermodynamic driving force, intrinsic diffusion coefficient; Phase transformations: spinodal decomposition, phase separation, order-disordered transformation, nucleation, martensitic transformations; Segregation: the driving force of segregation, isotherms; Interfaces, grain boundaries; Grain-boundary diffusion: Harrison's classification; Diffraction, reflectometry: diffraction theory of one-dimensional periodic structures in direct and reciprocal space, neutron and X-ray diffraction / reflectometry; Domain magnetism.		
Literature		
C.Kittel: Introduction to Solid State Physics William D. Callister, Jr. David G. Rethwisch Materials Science and Engineering, An Introduction, Willey M.A. Omar: Elementary Solid State Physics, Priciples and Applications CALPHAD (Calculation of Phase Diagrams): A Comprehensive Guide, Volume 1, 1st Edition, Editors: N. Saunders A.P. Miodownik		
Schedule:		
1 <sup>st</sup> week		
Information, introduction.		
Phase diagrams: Basic thermodynamic description of binary systems		
2 <sup>na</sup> week		
Phase Diagrams: Basics of calculations of real phase diagrams, Basics of the Calphad method.		

 $3^{rd}$  week Diffusion under external driving forces: Nernst-Einstein equation, thermodynamic driving force, intrinsic diffusion coefficient

 $4^{th}$  week

Phase transformations: spinodal decomposition ("up-hill" diffusion), phase separation ("down-hill" diffusion)

#### $5^{th}$ week

Phase transformations: order-disorder transformation

### $6^{th}$ week

Phase transformation: nucleation

#### 7<sup>th</sup> week

Phase transitions: martensite transformations, shape memory materials

### $8^{th}$ week

Segregation: Understanding driving forces of the segregation, segregation isotherms

### 9<sup>th</sup> week

Understanding the structure of interfaces and grain boundaries; grain-boundary diffusion: Harrison's classification of grain boundaries.

## $10^{th}$ week

Diffraction, reflectometry: diffraction theory of one-dimensional periodic structures in direct and reciprocal space, neutron and X-ray diffraction / reflectometry; Measurements in symmetrical geometry.

## 11<sup>th</sup> week

Diffraction, reflectometry: diffraction theory of one-dimensional periodic structures in direct and reciprocal space, neutron and X-ray diffraction / reflectometry;

 $12^{th}$  week

Domain magnetism: The basic idea of domain magnetism, domain structure.

13<sup>th</sup> week

Domain magnetism: Interaction of magnetic domains, domain reordering.

14<sup>th</sup> week

Summary, consultation

## **Requirements:**

- for a signature

Attendance at **lectures** is recommended, but not compulsory.

Participation at **practice classes** is compulsory. A student must attend the practice classes and may not miss more than three times during the semester. In case a student does so, the subject will not be signed and the student must repeat the course.

During the semester there are tests. The requirement for a signature is a successful (> 50%) completion of each test.

### - for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have a practical signature.

The grade for the examination is given according to the following table:

Score	Grade
0-49	fail (1)
50-62	pass (2)
63-75	satisfactory (3)
76-88	good (4)
89-100	excellent (5)

## Person responsible for course: Prof. Dr. Zoltán Erdélyi, full professor, DSc

Lecturer: Prof. Dr. Zoltán Erdélyi, full professor, DSc

<b>Title of course</b> : Theory of relativity <b>Code</b> : TTFME114	ECTS Credit points: 5
Type of teaching, contact hours	
- lectures: 2 hours/week	
- problem classes: 1 + 1 hours/week	
- laboratory: -	
Evaluation: signature + colloquium	
Workload (estimated), divided into contact hours:	
- lectures: 28 hours	
- problem classes: 14 + 14 hours	
- laboratory: -	
- home assignment: 40 hours	
- preparation for the exam: 40 hours	
Total: 136 hours	
<b>Year, semester</b> : 1 <sup>st</sup> year, 2 <sup>nd</sup> semester	
Its prerequisite(s): -	
Further courses built on it: -	

#### **Topics of course**

Physics and geometry, frames of reference, coordinate systems. Relativity principle in classical mechanics and electrodynamics. Propagation of light. Coordinate transformations, Minkowski space. Tensors. The covariance of physical laws. Covariant formulation of electrodynamics. Relativistic mechanics; relativistic collisions and phase space. Elements of relativistic field theory (the principle of least action, field equations). The electromagnetic field, gauge invariance, charge conservation. Symmetries and conservation laws, stress-energy tensor.

Gravitation, gravitational and inertial mass. The equivalence principle. General coordinate systems, geometric concepts (parallel transport, curvature tensor, etc.). Electrodynamics in curvilinear coordinate systems. Mechanics, inertial forces. Einstein equations, simple solutions (spherically symmetric solution, Schwarzschild metric, black holes). The Robertson-Walker metric and Friedmann cosmological models. Linearization of Einstein equations. Wave solutions, gravitational waves and their detection.

#### Literature

Compulsory:

Taylor, E.F., Wheeler, J.A.: Spacetime Physics, opensource 1992
Landau, L.D., Lifshitz, E.M.: The Classical Theory of Fields, Pergamon Press 1971
Susskind, L. and Friedman, A.: Special relativity and Classical Field Theory, 2017 *Recommended:*R.M. Wald, R.M.: General Relativity, The University of Chicago Press, 1984.
Misner, C.W., Thorne, K.S., Wheeler, J.A.: Gravitation, W. H. Freeman, 1973.
E. Byckling, K. Kajantie: Particle Kinematics, Wiley-Interscience, 1973

#### Schedule:

1<sup>st</sup> week

Geometry and coordinate systems. Coordinate transformations, covariant and contravari-ant coordinates. Galilei's principle of relativity (Galilei transformation and Newton axi-oms).
The propagation and speed of light (electrodynamics and Galilei transformation, Michel-son-Morley types of experiments).

 $2^{nd}$  week

Basic assumptions of the theory of relativity. Line element, invariant length. Lorentz trans-formation and its consequences (addition of velocities, length contraction, time dilation, experiments).

 $3^{rd}$  week

The event space. The geometry of Minkowski space, spacetime diagrams. Tensors. Opera-tions with tensors, tensor analysis. The covariance of physical laws.

 $4^{th}$  week

The covariant form of Maxwell equations. Four-potentials, electromagnetic field-strength tensor. Kinematics of a point particle. Proper time. The twin paradox and its interpretation. Four-velocity and acceleration.

 $5^{th}$  week

Fundamentals of the relativistic dynamics a point particle (relativistic momentum, energy, equation of motion). Relativistic collisions, phase space. The variation principle in a co-variant form. Motion of a point mass in a force field (charged particle in electromagnetic field, Lagrangian, Lorentz force).  $6^{th}$  week

Fundamentals of relativistic field theory. Hamilton principle, Euler-Lagrange equations. Applying the general description in the description of electromagnetic fields.

7<sup>th</sup> week

Space time symmetries in field theory. Stress-energy tensor, conservation laws.

 $8^{th}$  week

Basics of the theory of general relativity. Gravitational and inertial mass. The equivalence principle, local inertial systems.

9<sup>th</sup> week

Curvilinear coordinate systems (tensors, tensor analysis, parallel transport, covariant de-rivative). Christoffel symbols, metric tensor.

 $10^{th}$  week

Electrodynamics and mechanics in curvilinear coordinate systems. Geometry of spacetime and inertial forces.

 $11^{th}$  week

Curvature tensor, Ricci-tensor. The relationship between geometry and matter, Einstein equations.  $12^{th}$  week

Solving the Einstein equations for simple systems. Gravitational field of a spherically symmetric star. Schwarzschild metric, black holes. Robertson-Walker metric, Friedmann cosmological models. 13<sup>th</sup> week

The linearized Einstein equations. Wave solutions, gravitational waves and their detection.

14<sup>th</sup> week

Overview of the semester's material, guidance for further studies, consultation.

**Requirements:** 

- for a signature

Attendance at **lectures** is recommended, but not compulsory.

Participation at **problem classes** is compulsory. A student must attend the problem classes and may not miss more than three times during the semester. In case a student does so, the subject will not be signed and the student must repeat the course.

During the course, problems given for homework should be evaluated and submitted by a specified deadline. The requirement for a signature is a successful (> 50%) completion of the home assignments.

- for a grade

The course ends in an oral **examination/colloquium**. The requirement for applying for the exam is to have a signature.

The grade of the exam is given as follows.

Knowledge of the laws, theorems and definitions relevant to the subject : pass (2)

-in addition, the ability to prove the most important theorems : satisfactory (3)

-in addition, knowledge of proofs and derivations presented at the lectures : good (4)

-in addition, knowledge of the applications involved in the problem classes : excellent (5)

Person responsible for course: Dr. Schram Zsolt, habil. associate professor, PhD, CSc

Lecturer:

<b>Title of course</b> : Quantum Field Theory <b>Code</b> : TTFME0215	ECTS Credit points: 6
Type of teaching, contact hours	
- lecture: 2 hours/week	
- practice: 1 hours/week	
- laboratory: -	
Evaluation: exam	
Workload (estimated), divided into contact hours:	
- lecture: 28 hours	
- practice: 14 hours	
- laboratory: -	
- home assignment: 40 hours	
- preparation for the exam: 40 hours	
Total: 122 hours	
Year, semester: 1 <sup>st</sup> year, 2 <sup>nd</sup> semester	
Its prerequisite(s): Quantum Mechanics, Relativity theory	
Further courses built on it: Standard Model	
Topics of course	
Attempts at a relativistic quantum mechanics, the Klein-Gor appearance of quantized fields; classical field theory, canor physical significance and computation of the scattering c	rdon equation, the Dirac equation, the nical quantization of scalar fields; the ross section and the decay rate; the

appearance of quantized fields; classical field theory, canonical quantization of scalar fields; the physical significance and computation of the scattering cross section and the decay rate; the scattering amplitude and the LSZ reduction formula; path integral for free and interacting fields; perturbation theory, Feynman graphs, Feynman rules; Lagrangian density for fermion fields, the canonical quantization of the Dirac field; photons and quantum electrodynamics; calculations of cross sections decay widths for elementary processes.

# Literature

Compulsory:

Srednicki, M.: Quantum Field Theory, Cambridge University Press, 2007. *Recommended:* 

Peskin, M.E., Schroeder, D.V.: An Introduction to Quantum Field Theory, Westview Press, 1995. Mandl, F., Shaw, G.: Quantum Field Theory, Wiley, 1984.

Weinberg, S.: The Quantum Theory of Fields, Volume I, Cambridge University Press, 1995.

# Schedule:

1<sup>st</sup> week

Attempts at a relativistic quantum mechanics I: the abstract Schrödinger equation; Lorentz transformation; first attempt: the Klein-Gordon equation; second attempt: the Dirac equation.  $2^{nd}$  week

Attempts at a relativistic quantum mechanics II: third attempt: the introduction of fields; rewriting the quantum theory of non-relativistic bosons as a non-relativistic field theory.

 $3^{rd}$  week

Canonical quantization of scalar fields: the quantum mechanical Hamiltonian of free, relativistic bosons; Lagrangian for the Klein-Gordon equation, the Hamilton function of a free, classical scalar field; canonical quantization; the quantum mechanical Hamiltonian of free, relativistic bosons = the

Hamiltonian of the free, canonically quantized Klein-Gordon scalar field; the relationship between spin and statistics.

4<sup>th</sup> week

Scattering theory, LSZ reduction, n-point correlation functions: the cross section and decay rate; ``in'' and ``out'' states, the S-matrix; the LSZ reduction formula; n-point correlation functions; time-ordered product of operators; the normalization of fields.

# $5^{th}$ week

Path integral in quantum mechanics: general form of the path integral; the path integral in Lagrangian form; vacuum-to-vacuum transition amplitudes.

 $6^{th}$  week

Path integral in quantum field theory I: path integral for free scalar fields; the Feynman-propagator; Wick's theorem.

# $7^{th}$ week

Path integral in quantum field theory II: the path integral for interacting fields; Feynman-diagrams, Feynman-rules; counterterms.

# $8^{th}$ week

Scattering amplitudes, cross sections, decay rates: the computation of tree-level  $2 \rightarrow 2$  scattering amplitudes in scalar field theory; Feynman-rules in momentum space; computing cross sections at tree-level; decay rates and the LSZ reduction formula.

# 9<sup>th</sup> week

Canonical quantization of the Dirac field: the relativistic properties of the Dirac equation; the plane wave solutions of the Dirac equation; Lagrangian density for the Dirac equation; canonical quantization; the Hamilton function; the relationship of spin and statistics.

# $10^{th}$ week

Feynman-rules for fermion fields: correlation functions of fermion fields, spinors; propagator for the Dirac spinor; Yukawa theory, fermion-fermion-scalar vertex.

11<sup>th</sup> week

Feynman-rules for vector fields: Maxwell's equations; gauge invariance; photon polarization, photon propagator; Feynman-rules for QED; gauge invariance in the language of Feynman-diagrams.

 $12^{th}$  week

Elementary processes I: the square of the scattering amplitude; summation for spins and polarizations; spinor and gamma matrix technology; scattering amplitude.

 $13^{th}$  week

Elementary processes II: calculation of cross sections and decay rates in QED and Yukawa theory for  $2 \rightarrow 2$  and  $1 \rightarrow 3$  processes.

14<sup>th</sup> week

Review of the topics covered during the semester, guidance for further study, consultation.

# **Requirements:**

- for a signature

Being present on at least 75% of the practicals and correctly solving at least half of the assigned homework problems are prerequisites for taking the final exam.

- for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have a practical signature. The grade for the examination is given according to the following:

knowledge of the definitions, theorems and laws of the subject matter at the exam: pass (2); above this the ability to prove the more important theorems: satisfactory (3);

above this the knowledge of proofs and derivations mentioned in the lectures: good (4); above this the knowledge of the applications studied in the practicals: excellent (5).

Person responsible for course: Dr. Ádám Kardos

Lecturer: Dr. Ádám Kardos

<b>Title of course</b> : Particle Physics 2 <b>Code</b> : TTFME0112	ECTS Credit points: 4
Type of teaching, contact hours	
- lecture: 2 hours/week	
- practice: 1 hours/week	
- laboratory: -	
Evaluation: exam	
Workload (estimated), divided into contact hours:	
- lecture: 28 hours	
- practice: 14 hours	
- laboratory: -	
- home assignment: 40 hours	
- preparation for the exam: 40 hours	
Total: 122 hours	
<b>Year, semester</b> : 2 <sup>nd</sup> year, 1 <sup>st</sup> semester	
Its prerequisite(s): Particle Physics 1 (TFME0104)	
<b>Further courses built on it</b> : Quantum Mechanics 2 (TFME0102) Rela	tivity Theory (TFME0114)
Topics of course	
Gauge symmetries and interactions: QED, QCD and electroweak is Standard Model. Computer analysis of experimental data. Experimental Search for and observation of the Higgs boson. Results of LEP, H Problems and extensions of the Standard Model. Heavy ion physics.	nteraction. Structure of the l tests of the standard model. IERA, Tevatron and LHC.
Literature	
<i>Compulsory:</i> Dezső Horváth and Zoltán Trócsányi: Intorduction to Particle Ph Publishing, 2019. <i>Recommended:</i> Donald H. Perkins: Introduction to High Energy Physics, Addison-We Francis Halzen and Alan D.~Martin: Quarks and Leptons. An Intro Particle Physics, John Wiley and Sons, New York.	ysics, Cambridge Scholars esley, Menlo Park, USA oductory Course in Modern
<b>Schedule:</b> $1^{st}$ week Overview of the course Particle Physics 1 and the structure of the prese $2^{nd}$ week	ent course.

Global and local gauge symmetries. Local U(1) symmetry and electromagnetism. Perturbation calculation, scattering of point charges. Photon as gauge boson. Invariant (Mandelstam) variables. Feynman diagrams, photon reactions.

3<sup>rd</sup> week

Local SU(3) symmetry and strong interaction. Running coupling constants. Quark confinement, asymptotic freedom. Parton model and hadron jets. Event shape quantities: thrust, sphericity, etc. Gluonia: their spin and colour charge.

4<sup>th</sup> week

Parity violation in beta decay, V-A theory. Local SU(2) symmetry does not yield weak interaction.. Spontaneous symmetry breaking, the Brout-Englert-Higgs mechanism. Weak isospin. Parity violation. Weak current and coupling, neutral current. Weinberg angle.  $5^{th}$  week

Mass production in the BEH mechanism. Weak boson and fermion masses. GIM mechanism. Mixing of quark states. CP violation. Cabibbo-Kobayashi-Maskawa matrix. The Lagrangian of the Standard Model.

# $6^{th}$ week

Steps of data analysis. Role of simulations. Covariance, correlation, standard deviation, error propagation. Parameter estimation and its quality, confidence. Trigger logic.

 $7^{th}$  week

Linear and nonlinear parameter estimation: ROOT and MINUIT program packages. Statistic and systematic uncertainties. Discovery and exclusion. Estimating upper and lower limits. Rules of thumb.

 $8^{th}$  week

Direct and indirect experimental proofs of SM. Parametrization and fitting the SM. Forwardbackward and right-left asymmetries. Lepton universality. W physics. Tetra-, penta- and hexaquarks.

9<sup>th</sup> week

Basic experiments: Proton-proton collisions, CDF and the top quark. Particle searches with CMS at the LHC.

10<sup>th</sup> week

Higgs boson of the Standard Model. Its possible production and decay. Higgs search: LEP, Tevatron. Its observation at the LHC. The 2-Higgs-doublet models and its search.

 $11^{th}$  week

Problem of the Standard Model. Convergence of gauge couplings. Grand unification. Hierarchy problem and its possible solution.

 $12^{th}$  week

Supersymmetry and the Minimal Supersymmetric Standard Model (MSSM). Search for MSSM-particles.

 $13^{th}$  week

Quark-gluon plasma and its signals. Heavy ion experiments: SPS NA61/SHINE, RHIC PHENIX and LHC ALICE. Hadron jet quenching. Heavy-ion results at the LHC.

14<sup>th</sup> week

Overview, discussion session.

# **Requirements:**

- for a signature

Attendance at **lectures** is recommended, but not compulsory.

Participation at **practice classes** is compulsory. A student must attend the practice classes and may not miss more than three times during the semester. In case a student does so, the subject will not be signed and the student must repeat the course.

During the semester there are practical home works. The requirement for a signature is a successful (> 50%) completion of the home works.

- for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have a practical signature. The examination starts with a quiz: 8-12 questions with multiple choice answers. The students should be able to explain the chosen answers.

Scoring above 50%: *satisfactory*, above 75%: *average*, in addition, the explanations are correct as well: *good*, in addition, the interpretation of the knowledge is correct: *excellent*.

Person responsible for course: Dr. Zoltán Trócsányi, professor, PhD

Lecturer: Dr. Dezső Horváth, professor, PhD

<b>Title of course</b> : The Standard Model of Particle Physics <b>Code</b> : TTFME0111	ECTS Credit points: 4	
Type of teaching, contact hours		
- lecture: 2 hours/week		
- practice: 1 hours/week		
- laboratory: -		
Evaluation: exam		
Workload (estimated), divided into contact hours:		
- lecture: 28 hours		
- practice: 14 hours		
- laboratory: -		
- home assignment: 40 hours		
- preparation for the exam: 40 hours		
Total: 122 hours		
Year, semester: 2 <sup>nd</sup> year, 1 <sup>st</sup> semester		
Its prerequisite(s): Quantum Field Theory (TTFME0113)		
Further courses built on it: -		
Topics of course		
Wigner theorem, Lagrangian for massive and massless spin-1 particles. Generalisation of Lorentz transformations, spinors, Lagrangian for spin-1/2 fermions, Lagrangian of quantum electrodynamics, Feynman rules of QED, Rutherford scattering, muon-pair production in electron-positron annihilation. Spontaneous breaking of discrete, global and local symmetries. Abelian and non-Abelian Higgs models. Electroweak symmetry breaking. The fermion sector, fermion masses and mixing angles.		
Literature		
<i>Compulsory:</i> M. Schwartz, Quantum Field Theory and the Standard Model (Cambridge University Press, 2013)		
Schedule:		
1 <sup>st</sup> week		
Unitary representation of the Poincare group, Wigner theorem, embedding	ing spin-0 particles into fields.	
$2^{nd}$ week		
Embedding massive and massless spin-1 particles into fields.		
Covariant derivatives and scalar OED quantization of messive and r	needlass onin 1 fields and the	
Ward identity.	hassiess spin-1 heids and the	
$4^m$ week		
Group theory and group representations of the Lorentz group. Spinor group, Weyl spinors.	representations of the Lorentz	

 $5^{th}$  week

Constructing Lorentz invariants Lagrangians with spinors, gamma matrices, coupling to photons.  $6^{th}$  week

Spinor solutions to the Dirac equations.

7<sup>th</sup> week

Quantizing spinors, invariance of S-matrix, Feynman propagator for spin-1/2 particles.

 $8^{th}$  week

Quantum electrodynamics, Feynman rules, signs, gamma matrix identities.

9<sup>th</sup> week

Rutherford scattering, muon-pair production.

10<sup>th</sup> week

Spontaneous breaking of discrete, global and local symmetries. Goldstone bosons and the Goldstone theorem.

 $11^{th}$  week

Abelian Higgs model, unitary gauge and Higgs mechanism.

 $12^{th}$  week

Electroweak symmetry breaking, masses of vector bosons of the Standard model.

 $13^{th}$  week

Fermion sector, mass generation in lepton and quark sectors.

14<sup>th</sup> week

Fermion masses and mixing angles. Mass and flavor basis, the CKM matrix.

**Requirements:** 

- for a signature

Attendance at **lectures** is recommended, but not compulsory.

Participation at **practice classes** is compulsory. A student must attend the practice classes and may not miss more than three times during the semester. In case a student does so, the subject will not be signed and the student must repeat the course.

During the semester there are practical home works which have to be evaluated and submitted by the end of the  $14^{\text{th}}$  week of the semester. The requirement for a signature is a successful (> 50%) completion of the home works.

- for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have a practical signature.

The grade for the examination is given according to the following table:

Score	Grade
0-49	fail (1)
50-62	pass (2)
63-75	satisfactory (3)
76-88	good (4)
89-100	excellent (5)

Person responsible for course: Prof. Zoltán Trócsányi

Lecturer: Dr. Adam Kardos, assistant professor, PhD

<b>Title of course</b> : Electronic structure methods and quantum dynamics <b>Code</b> : TTFME0121	ECTS Credit points: 5
Type of teaching, contact hours	
- lecture: 2 hours/week	
- practice: 1 hours/week	
- laboratory: -	
Evaluation: exam	
Workload (estimated), divided into contact hours:	
- lecture: 28 hours	
- practice: 14 hours	
- laboratory: -	
- home assignment: 40 hours	
- preparation for the exam: 40 hours	
Total: 122 hours	
<b>Year, semester</b> : 1 <sup>st</sup> year, 2 <sup>st</sup> semester	
Its prerequisite(s): -	
Further courses built on it: -	
Tomics of course	

#### **Topics of course**

Schrödinger equation of atoms and molecules. Center of mass motion. The Born-Oppenheimer and adiabatic approximation. Atomic units. The differential Hellmann-Feynman theorem. The integral Hellmann-Feynman theorem. The virial theorem. The variation principle. Variational methods. The linear variation method (Ritz method). Eckart's inequality. Scaling - a connection with the variation principle. Perturbational methods. Non-degenerate Rayleigh-Schrödinger perturbation theory. Degenerate Rayleigh-Schrödinger perturbation theory. Wave functions. Spin orbitals. Antisymmetric wave functions. Singlet and triplet states. Slater determinants.

Determinant wave functions. Matrix elements between determinant wave functions. The Hartree-Fock method. The self consistent field (SCF) approximation. The configuration interaction (CI) method. Koopmans theorem. Singlet and triplet excitations. Electron correlation. The multi reference configuration interaction MRCI and multi reference self-consistent field (MCSCF) approximations. The coupled cluster approach. The Schrödinger equations of the nuclei. Vibrational eigenstates. Time-dependent perturbation theory (first and second order). Time evolution of non-stationary systems. Methods for propagating the

nuclear wave packet: finite difference method, split operator technique, the relation between coordinate and momentum representations. Analysis of the time-dependent nuclear wave packet: initial states, autocorrelation functions, populations of electronic states, complex absorbing potentials, dissociation of molecules. Further propagation techniques: time-dependent Hartree and the multiconfigurational time-dependent Hartree methods. Structure of the total wave function, basis functions. Time evolution on multiple electronic states, non-adiabatic processes.

#### Literature

D. R. Yarkony: Modern Electronic Structure Theory, World Scientific, 1995. I. Mayer: Simple Theorems, Proofs, and Derivations in Quantum Chemistry, Kluwer Academic, 2003.

# Schedule:

1<sup>st</sup> week

Schrödinger equation of atoms and molecules. Center of mass motion.

 $2^{nd}$  week

The Born-Oppenheimer and adiabatic approximation.

3<sup>rd</sup> week

Atomic units. The differential Hellmann-Feynman theorem. The integral Hellmann-Feynman theorem. The virial theorem.

4<sup>th</sup> week

The variation principle. Variational methods. The linear variation method (Ritz method).

 $5^{th}$  week

Eckart's inequality. Scaling - a connection with the variation principle. Perturbational methods. Nondegenerate Rayleigh-Schrödinger perturbation theory.

 $6^{th}$  week

Degenerate Rayleigh-Schrödinger perturbation theory. Brillouin-Wigner perturbation theory. 7<sup>th</sup> week

Wave functions. Spin orbitals. Antisymmetric wave functions. Singlet and triplet states. Slater determinants.

 $8^{th}$  week

Determinant wave functions. Matrix elements between determinant wave functions.

9<sup>th</sup> week

The Hartree-Fock method. The self consistent field (SCF) approximation. The configuration interaction (CI) method.

 $10^{th}$  week

Koopmans theorem. Singlet and triplet excitations. Electron correlation. The multi reference configuration interaction MRCI and multi reference self-consistent field (MCSCF) approximations. The coupled cluster approach.

 $11^{th}$  week

The Schrödinger equations of the nuclei. Vibrational eigenstates. Time-dependent perturbation theory (first and second order). Time evolution of non-stationary systems.

 $12^{th}$  week

Methods for propagating the nuclear wave packet: finite difference method, split operator technique, the relation between coordinate and momentum representations.

 $13^{th}$  week

Analysis of the time-dependent nuclear wave packet: initial states, autocorrelation functions, populations of electronic states, complex absorbing potentials, dissociation of molecules.

14<sup>th</sup> week

Further propagation techniques: time-dependent Hartree and the multiconfigurational time-dependent Hartree methods. Structure of the total wave function, basis functions. Time evolution on multiple electronic states, non-adiabatic processes.Addition of angular momenta, the Clebsch-Gordan coefficients.

# **Requirements:**

- for a signature

Attendance at **lectures** is recommended, but not compulsory.

Participation at **practice classes** is compulsory. A student must attend the practice classes and may not miss more than three times during the semester. In case a student does so, the subject will not be signed and the student must repeat the course.

During the semester there are practical home works which have to be evaluated and submitted by the		
end of the 14 <sup>th</sup> week of the semester. The requirement for a signature is a successful (> 50%)		
completion of the home works.		
- for a grade		
The course ends in an examin	ation. The requirement for applying for an exam is to have a practical	
signature.		
The grade for the examination is given according to the following table:		
Score	Grade	
0-49	fail (1)	
50-62	pass (2)	
63-75	satisfactory (3)	
76-88	good (4)	
89-100	excellent (5)	
Person responsible for course: Prof. Dr. Ágnes Vibók, PhD, habil, DSc		
Lecturer: Prof. Dr. Ágnes Vibók, PhD, habil, DSc		

<b>Title of course</b> : Atoms and molecules in electromagnetic field <b>Code</b> : TTFME0122	ECTS Credit points: 5
Type of teaching, contact hours	
- lecture: 2 hours/week	
- practice: 1 hours/week	
- laboratory: -	
Evaluation: exam	
Workload (estimated), divided into contact hours:	
- lecture: 28 hours	
- practice: 14 hours	
- laboratory: -	
- home assignment: 40 hours	
- preparation for the exam: 40 hours	
Total: 122 hours	
<b>Year, semester</b> : 2 <sup>st</sup> year, 1 <sup>st</sup> semester	
Its prerequisite(s): -	
Further courses built on it: -	
Topics of course	

Two and three level atoms. Spin matrices. Properties of spin matrices. Ladder operators for two and three level atoms. Linear harmonic oscillator. Algebra of ladder operators. Matrix

representation. Introduction to the quantum optics. Quantization of electromagnetic field. Coherent and squeezed states. Angular momentum algebra. Eigenvalues and eigenvectors of angular momentum and spin operators. Composite systems. Two dimensional isotropic oscillator. Algebra of ladder operators. Three dimensional isotropic oscillator. Eigenvectors and eigenvalues. The angular momentum algebra. Clebsch-Gordan coefficients. Triplet and singlet states of spin and their algebra. Isospin and isomultiplets. Time-dependent Hamiltonian. Perturbative series. First and second order corrections. Energy eigenvalues and eigenfunctions. Harmonic and anharmonic perturbations. Perturbed oscillator. Light-matter interaction. Dipole approximation. Non-degenerate first-order perturbation correction. Absorption and emission. Perturbed hydrogen atom. Hydrogen atom in electric field. The linear Stark effect. Hydrogen atom in magnetic field.

The Zeeman effect. Interaction of a two-level atom with radiation. Dipole approximation. Rabi-model. The rotating wave approximation (RWA). The Jaynes-Cummings model. Atom optic. Spontaneous emission.

# Literature

V. K. Thankappan: Quantum Mechanics, Wiley Eastern Limited, 1985.

C. C. Gerry, P. L. Knight: Introductory Quantum Optics, Cambridge University press, 2005.

#### Schedule:

1<sup>st</sup> week

Two and three level atoms. Spin matrices. Properties of spin matrices. Angular momentum algebra. Ladder operators for two and three level atoms.

#### $2^{nd}$ week

Linear harmonic oscillator. Algebra of ladder operators. Matrix representation.

 $3^{rd}$  week

Introduction to the quantum optics. Quantization of electromagnetic field. Coherent and squeezed states.  $4^{th}$  week

4<sup>™</sup> week

Angular momentum algebra. Eigenvalues and eigenvectors of angular momentum and spin operators.  $5^{th}$  week

Two dimensional isotropic oscillator. Algebra of ladder operators. Three dimensional isotropic oscillator. Eigenvectors and eigenvalues.

 $6^{th}$  week

The angular momentum algebra. Clebsch-Gordan coefficients.

 $7^{th}$  week

Triplet and singlet states of spin and their algebra. Isospin and isomultiplets.

 $8^{th}$  week

Time-dependent Hamiltonian. Perturbative series. First and

second order corrections. Energy eigenvalues and eigenfunctions.

Harmonic and anharmonic perturbations. Perturbed oscillator.

 $9^{th}$  week

Light-matter interaction. Dipole approximation. Non-degenerate

first-order perturbation correction.

 $10^{th}$  week

Absorption and emission. Perturbed hydrogen atom. Hydrogen

atom in electric field. The linear Stark effect.

11<sup>th</sup> week

Hydrogen atom in magnetic field. The Zeeman effect.

12<sup>th</sup> week

Interaction of a two-level atom with radiation. Dipole approximation. Rabi-model. The rotating wave approximation (RWA).

13<sup>th</sup> week

The Jaynes-Cummings model. Atom optic. Spontaneous emission.

14<sup>th</sup> week

Closed-term exam.

# **Requirements:**

- for a signature

Attendance at **lectures** is recommended, but not compulsory.

Participation at **practice classes** is compulsory. A student must attend the practice classes and may not miss more than three times during the semester. In case a student does so, the subject will not be signed and the student must repeat the course.

During the semester there are practical home works which have to be evaluated and submitted by the end of the  $14^{th}$  week of the semester. The requirement for a signature is a successful (> 50%) completion of the home works.

- for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have a practical signature.

The grade for the examination is given according to the following table: Score Grade

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Lecturer: Prof Dr. Ágnes Vibók, PhD, habil, DSc.		
Person responsible for course: Prof Dr Ágnes Vibók PhD habil DSc		
89-100	excellent (5)	
76-88	good (4)	
63-75	satisfactory (3)	
50-62	pass (2)	
0-49	fail (1)	

<b>Title of course</b> : Atomic and molecular physics II. <b>Code</b> : TTFME0125	ECTS Credit points: 5	
Type of teaching, contact hours		
- lecture: 2 hours/week		
- practice: 1 hours/week		
- laboratory: -		
Evaluation: exam		
Workload (estimated), divided into contact hours:		
- lecture: 28 hours		
- practice: 14 hours		
- laboratory: -		
- home assignment: 40 hours		
- preparation for the exam: 40 hours		
Total: 122 hours		
<b>Year, semester</b> : 1 <sup>st</sup> year, 2 <sup>nd</sup> semester		
Its prerequisite(s): Atomic and molecular physics I.		
Further courses built on it: -		
Topics of course		
Statistical description of atoms and molecules. Pure and mixed states. The density matrix. The Liouville-Neumann equation, Liouville operator. Environmental effects, the reduced density matrix. Classical scattering, basic concepts. Atomic collisions, partial waves, radial equations. Green-function. Contour integrals. Born series. Scattering by a complex potential. Electron-atom collisions, static-exchange and close coupling methods. Optical potentials. Atomic excitation by collision. Ionization and resonances. Atom-atom and atom-molecule collisions at low velocities. Electronic excitations and charge transfer.		
Literature		
Compulsory: <b>B</b> H Bransdan, C. I. Josephsin: Physics of stoms and molecules. Longm	an Scientific & Technical	

B. H. Bransden, C. J. Joachain: Physics of atoms and molecules, Longman Scientific & Technical B. W. Shore: The theory of coherent atomic excitation, JOHN WILEY & SONS *Recommended*:

D. J. Griffiths: Introduction to Quantum Mechanics, Prentice-Hall: New Jersey

# Schedule:

1<sup>st</sup> week

Statistical description of atomic and molecular systems. Statistical nature of initial states: pure and mixed states. Coherent superposition.

 $2^{nd}$  week

Density matrix and its fundamental properties. Operator expectation values for pure and mixed states. Incoherent superposition. Thermal states.

 $3^{rd}$  week

The Liouville-Neumann equation. Constants of motion, probability amplitudes, and coherences. The Liouville operator. Applications.

 $4^{th}$  week

Atomic and molecular systems interacting with the environment. The reduced density matrix. Relaxations, fluctuations and scattering processes.

 $5^{th}$  week

Classical scattering on a central potential, impact parameter, differential and total cross sections. Elastic scattering on a rigid sphere. Coulomb scattering.

 $6^{th}$  week

Atomic collisions. Basic concepts and the potential scattering. Types of collisions, channels and thresholds.

 $7^{th}$  week

Scattering amplitude, the method of partial waves. Radial equations, Bessel equation. Hankel and Bessel functions. Phase shifts and resonances.

 $8^{th}$  week

Integral equation of potantial scattering and its solution by Green's function method. Contour integrals.  $9^{th}$  week

Born series expansion of the scattering amplitude. The first Born approximation for Coulomb potential, relation to the classical results. Absorption and scattering by a complex potential.

#### $10^{th}$ week

Electron-atom collisions. Nonrelativistic electron scattering on atomic hidrogen. Modelling elastic scattering: static exchange and close coupling methods.

11<sup>th</sup> week

Excitation of atoms to discrete levels. Ionization and autoionization processes. Resonances. The electron energy loss spectrum.

 $12^{th}$  week

Atom-atom collisions. Long-range interaction, classical description. Elastic scattering at low velocities. Electronic excitation and charge exchange. Aspect of molecular collisions.

13<sup>th</sup> week

Summary and consultation.

*14<sup>th</sup> week* End-term test.

# **Requirements:**

- for a signature

Attendance at **lectures** is recommended, but not compulsory.

Participation at **practice classes** is compulsory. A student must attend the practice classes and may not miss more than three times during the semester. In case a student does so, the subject will not be signed and the student must repeat the course.

During the semester there are practical home works which have to be evaluated and submitted by the end of the  $14^{th}$  week of the semester. The requirement for a signature is a successful (> 50%) completion of the home works.

# - for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have a practical signature.

The grade for the examination is given according to the following table:

Score	Grade
0-49	fail (1)
50-62	pass (2)
63-75	satisfactory (3)

76-88 89-100	good (4) excellent (5)	
Person responsible for course: Dr. András Csehi, assistant professor, PhD		
Lecturer: Dr. András Csehi, assistant professor, PhD		

Title of course: Computer modelling Code: TTFME0132	ECTS Credit points: 5
Type of teaching, contact hours	
- lecture: 2 hours/week	
- practice: 1 hours/week	
- laboratory: -	
Evaluation: exam	
Workload (estimated), divided into contact hours:	
- lecture: 28 hours	
- practice: 14 hours	
- laboratory: -	
- home assignment: 40 hours	
- preparation for the exam: 40 hours	
Total: 122 hours	
Year, semester: 1 <sup>st</sup> year, 2 <sup>nd</sup> semester	
Its prerequisite(s): -	
Further courses built on it: -	
Topics of course	
Advanced methods of random number generation, Gaussian Generalization and optimization of the Hit&Miss method. Numerical integration by means of finite difference methods. Monte of sampling. Error analysis and optimization. MC integration with imp with Markov-chain algorithms, Markov-chain in the phase space of Statistical mechanics of finite temperature systems, size depend measurement, Metropolis algorithm, Metropolis and Glauber dynam Ising model in one and two dimensions. Numerical analysis of the Is Efficient implementation of the Metropolis algorithm for the Is conditions, optimization of the simulation program. Thermalizatia analysis of MC simulations. Monte Carlo simulation in the vicinity of the critical point. Critical Swendsen-Wang algorithm. Finite size scaling of simulated data, r exponents. Histogram methods. Simple histogram method of Ferrenberg. Broad Lattice gas models, conserved order parameter (COP) Ising model as COP Ising model. Kawasaki dynamics and its efficient implementat Metropolis algorithm for gases and liquids. Optimization of the M thermodynamic quantities of gases and liquids by means of MC simu Kinetic Monte Carlo (KMC) simulation. Poisson statistics of tran- waiting time. Residence time algorithm. Efficient implementation of	n distributed random numbers. Carlo (MC) integration with simple portance sampling. MC integration physical systems. ence of energy fluctuations. MC nics. ing model with exact enumeration. sing model. Initial and boundary on and MC measurements. Error slowing down, cluster algorithms, numerical determination of critical d histogram methods. a lattice gas. Phase diagram of the ion. C simulation program. Measuring ulations. asitions, transition probability and f KMC simulations.
Literature	
Compulsory:	

Kun Ferenc, Computer modeling and simulation in physics, (electronic script in English, University of Debrecen, 2012).

H. Gould and J. Tobochnik, An introduction to computer simulation methods (Addison-Wesley, 2006).

M. E. J. Newman and G. T. Barkema, Monte Carlo Methods in Statistical Physics (Oxford University Press, 1999).

# Schedule:

1<sup>st</sup> week

Advanced methods of random number generation. Hit&Miss method, error analysis and optimization.  $2^{nd}$  week

Generating Gaussian distributed random numbers, Box-Müller algorithm. Central limit theorem. Algorithms to generate random points on the surface and inside the volume of a sphere. Multidimensional random vectors.

# 3<sup>rd</sup> week

Numerical integration by means of finite difference methods, error analysis. Monte Carlo integration with simple sampling.

# $4^{th}$ week

Error analysis of Monte Carlo integration, optimization of MC integration. Monte Carlo integration with importance sampling, optimal choice of sampling functions.

# $5^{th}$ week

Monte Carlo integration with Markov-chain methods. Markov-chain in the phase space. Generating random numbers with Markov-chain algorithm.

 $6^{th}$  week

Statistical mechanics of finite temperature systems. Distribution of the energy of a finite temperature system, energy fluctuations and their dependence on the system size.

7<sup>th</sup> week

Generating sequences of micro-states with Markov-chain algorithms. Condition of detailed balance. Metropolis algorithm, Metropolis and Glauber dynamics.

 $8^{th}$  week

Ising model in one and two dimensions. Metropolis algorithm for the Ising model. Initial and boundary conditions, thermalization of the system. Optimization of the simulation program. Measuring thermodynamic quantities by means of Monte Carlo simulations.

9<sup>th</sup> week

Monte Carlo simulation in the vicinity of the critical point, numerical determination of critical exponents. Finite size scaling.

 $10^{th}$  week

Critical slowing down. Cluster algorithms and their efficient implementation in the Ising model. The Swendsen-Wang algorithm.

11<sup>th</sup> week

Foundation of histogram methods. Ferrenberg's simple histogram method. Monte Carlo simulation over a broad range of temperature, the broad histogram method.

 $12^{th}$  week

Lattice gas models, conserved order parameter (COP) Ising model as a lattice gas. Kawasaki dynamics and its efficient implementation for the COP Ising model.

 $13^{th}$  week

Metropolis algorithm for gases and liquids. Optimization of the MC simulation program. Measuring thermodynamic quantities of gases and liquids by means of MC simulations.

14<sup>th</sup> week

Kinetic Monte Carlo simulation. Poisson statistics of transitions, transition probability and waiting time. Residence time algorithm. Efficient implementation of kinetic Monte Carlo simulations.

#### **Requirements:**

- for a signature

Attendance at **lectures** is recommended, but not compulsory.

Participation at **practice classes** is compulsory. A student must attend the practice classes and may not miss more than three times during the semester. In case a student does so, the subject will not be signed and the student must repeat the course.

During the semester there are practical home works which have to be evaluated and submitted by the end of the 14<sup>th</sup> week of the semester. The requirement for a signature is a successful completion of all the home works.

# - for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have a practical signature.

The grade for the examination is given according to the following table:

Score	Grade
0-49	fail (1)
50-62	pass (2)
63-75	satisfactory (3)
76-88	good (4)
89-100	excellent (5)

Person responsible for course: Dr. Ferenc Kun, professor, PhD

Lecturer: Dr. Ferenc Kun, professor, PhD

<b>Title of course</b> : Complex networks and their applications <b>Code</b> : TTFME0134	ECTS Credit points: 5	
Type of teaching, contact hours		
- lecture: 2 hours/week		
- practice: 1 hours/week		
- laboratory: -		
Evaluation: exam		
Workload (estimated), divided into contact hours:		
- lecture: 28 hours		
- practice: 14 hours		
- laboratory: -		
- home assignment: 40 hours		
- preparation for the exam: 40 hours		
Total: 122 hours		
<b>Year, semester</b> : 2 <sup>nd</sup> year, 1 <sup>st</sup> semester		
Its prerequisite(s): -		
Further courses built on it: -		
Topics of course		
Networks in nature and society. Classification of networks.		
Network as a graph. The basics of graph theory. Directed and undirected	ed graphs. Node degree, average	
degree, degree distribution. Adjacency matrix. Weighted networks. P	Path and distance, shortest path,	
diameter, average path. Connectivity. Clustering coefficient.		
Random networks. The Erdős-Rényi model. Small world networks, six-step theory. Watts-Strogatz rewiring algorithm.		
Scale free networks. The Price model, Barabási-Albert model. Growth and preferential attachment. Extensions of the Barabási-Albert model. Universality.		
Generating networks with arbitrary degree distribution. The configuration model. The hidden parameter model. Non-linear preferential attachment. Origin of preferential attachment.		
Correlated networks. Robustness of networks.		
Percolation of networks. Inverse percolation transition. The Molloy-Reed criterion. Robustness against attack. Failure avalanches and their modelling. Model of failure spreading. Designing robust networks.		
Communities on networks and their identification. Social and biological networks. Spreading		
Immunization. Forecasting epidemic outbreaks.		
Literature		
Compulsory: Barabási Albert László, Network Science (Cambridge University Press, 2016). M. E. J. Newman, Networks: An Introduction (Oxford University Press, Oxford, 2010).		
Schedule:		
1 <sup>st</sup> week		
Networks in nature and society, real life examples. The history of network science. Relation of networks to complex systems. The potential of network approach.		
2 <sup>rm</sup> week		

Basics of graph theory. Directed and undirected networks. Adjacency matrix. Weighted networks. Bipartit networks. The Metcalfe-law.

 $3^{rd}$  week

Structural measures of complex networks: path, distance, shortest path, diameter, average path. Connectivity and clustering. The clustering coefficient.

 $4^{th}$  week

Random networks: the Erdős-Rényi model. Degree distribution and clustering coefficient. Structural phases of random networks. Comparison to real networks.

 $5^{th}$  week

Small world networks, the six-step distance. The Bacon-number. Watts-Strogatz rewiring algorithm and its computer implementation.

 $6^{th}$  week

Scale free networks. Network centers (hubs). Ultra-small world property. The degree exponent.

7<sup>th</sup> week

Generating complex networks with an arbitrary degree distribution. The configuration model. The hidden parameter model. Network diameter and clustering coefficient.

 $8^{th}$  week

The Price model and the Barabási-Albert model of scale free networks with directed and undirected links. Growth and preferential attachment. Degree dynamics, dynamic exponent, degree distribution. Diameter and clustering coefficient.

 $9^{th}$  week

Measuring the degree of preferential attachment. Non-linear preferential attachment. The origin of preferential attachment. Edge selection model, copying model.

 $10^{th}$  week

Evolving networks. The Bianconi-Barabási model. Fitness and fitness distribution. Measuring network fitness. Bose–Einstein-condensation. Extensions of the Barabási-Albert model. Removal of nodes. Accelerated growth and aging.

 $11^{th}$  week

Correlation of node degree, the degree correlation matrix. Measuring degree correlations. Degree correlation coefficient. Generation of correlated networks.

 $12^{th}$  week

Percolation on networks. Inverse percolation transition. The Molloy-Reed criterion. Robustness against attack. Failure avalanches and their modelling. Branching process, the branching model. Design of robust networks.

 $13^{th}$  week

Communities on social and biological networks. Klicks. Weak and strong communities. Decomposition of networks into communities. The Kerninghan-Lin algorithm. Hierarchical clustering. Integrating algorithms. The Ravasz algorithm. Dividing algorithms. The Girvan-Newman algorithm. Klick percolation.

14<sup>th</sup> week

Spreading phenomena. Modelling epidemic deseases. Network epidemics. Immunization. Forecasting of epidemic outbreaks.

# **Requirements:**

- for a signature

Attendance at **lectures** is recommended, but not compulsory.

Participation at **practice classes** is compulsory. A student must attend the practice classes and may not miss more than three times during the semester. In case a student does so, the subject will not be signed and the student must repeat the course.

During the semester there are practical home works which have to be evaluated and submitted by the end of the 14<sup>th</sup> week of the semester. The requirement for a signature is a successful completion of all the home works.

- for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have a practical signature.

The grade for the examination is given according to the following table:

Score	Grade
0-49	fail (1)
50-62	pass (2)
63-75	satisfactory (3)
76-88	good (4)
89-100	excellent (5)

Person responsible for course: Dr. Ferenc Kun, professor, PhD

Lecturer: Dr. Ferenc Kun, professor, PhD

<b>Title of course</b> : Phase Transitions and Critical Phenomena 1 <b>Code</b> : TTFME0133	ECTS Credit points: 5	
Type of teaching, contct hours		
- lecture: 2 hours/week		
- practice: 1 hours/week		
- laboratory: -		
Evaluation: exam		
Workload (estimated), divided into contact hours:		
- lecture: 28 hours		
- practice: 14 hours		
- laboratory: -		
- home assignment: 40 hours		
- preparation for the exam: 40 hours		
Total: 122 hours		
<b>Year, semester</b> : 1 <sup>st</sup> year, 1 <sup>st</sup> semester		
Its prerequisite(s): -		
Further courses built on it: -		
Topics of course		
Phase notion, order of the transition, continuous transitions, tran	sitions of	
n-order, Gibbs's phase rule, the effect of the dimensionality, role of		
fluctuations, effect of correlations, long range order, order param	neter,	
Ginsburg-Landau thermodynamic potential, Kadanoff block cor	nstruction,	
generalized homogenity, critical exponents, scaling laws, sever	e critical	
exponents and scaling laws, Orstein-Zernike behaviour, Landau	theory,	
mean-field critical exponents, universality classes, critical slowing-down,		
renormalization group transformation, critical surface, basic elements of the		
fixpoint theory, the relation between phase transitions and renormalization		
group, deduction of critical exponents via renormalization group theory.		
Literature		
Compulsory:		
N. Goldenfeld, Lectures on Phase Transitions and the Renormalization Group,		
Recommended:		
J.J. Binney, N. J. Dowrick, A. J. Fisher, M. E. J. Newman: The theory of		
critical phenomena and introduction to renormalization group, Oxford Science		
Publication, Clarendon Press, Oxford, 1995.		

# Schedule:

1<sup>st</sup> week

The phase notion, role of the thermodynamic potential, stable and metastable states, the order of the transition in the Ehrenfest's scheme, examples for 1-4 order transitions, continuous transitions, infinite order transitions, the Gibbs phase rule and its importance in phase diagrams.

 $2^{nd}$  week

The emergence possibility of phase transitions in 1D, Landau's deduction, notion of long range order. Essential parameters as: dimensionality, action range of the interactions, thermodynamic limit, number of components of the characteristique dynamical variable, the precence of T > 0 value, and their importance in qualitative changes inside the system.

 $3^{rd}$  week

The importance of fluctuations, their dimensionality dependence, the average of the square of local fluctuations. Long range order in two dimensions, the Mermin-Wagner theorem and its consequences.  $4^{th}$  week

The common breaking of continuous symmetries and gauge invariance. The average value of the correlations in geometrical space, its dimensionality dependence and its connection to fluctuations. The upper and lower critical dimensions, their meaning and role.

 $5^{th}$  week

The connection between continuous transitions and symmetry breaking, deduction of characteristics emerging from this connection. Expansion based on base functions of irreducible representations, critical regions and critical phenomena. The Landau order parameter and its connection to the symmetry breaking.  $6^{th}$  week

The Ginsburg-Landau thermodynamic potential for homogeneous and inhomogeneous systems, basic properties of the Ginsburg-Landau thermodynamic potential, spontaneous and non-spontaneous transitions. Characteristics and properties of the coefficients in the

Ginsburg-Landau thermodynamic potential.

 $7^{th}$  week

The correlation length and its critical exponent, Kadanoff block construction, block construction notion and characteristics, notion of block variables. Scale invariance and its consequences. Critical exponents, static scaling, generalised homogenity.

 $8^{th}$  week

Scaling of the correlation function, scaling laws connected to the thermodinamic potential and to the paircorrelation function, relations between critical exponents. Dimensionality in scaling laws, hyperscaling. Critical exponents and scaling laws in strict sense.

9<sup>th</sup> week

The Landau theory of phase transitions, the critical exponents of the Landau theory, mean-field critical exponents, universality classes. Evaluation of the Landau theory of phase transitions.

 $10^{th}$  week

*The Orstein-Zernike correlation behavior, direct short range correlations, entanglement notion and role. 11*<sup>th</sup> week

The renormalization group transformation, effectuation technique and characteristics. Flow in the parameter space of the Hamiltonian. The behavior at the phase transition, the flow far from phase transitions. Connection to universality classes.

 $12^{th}$  week

Fixpoints, their characteristics, the notion of the critical surface, the fixpoint notions in view of the renormalization group.

 $13^{th}$  week

Effectuation of the renormalization group transformation for different models.

14<sup>th</sup> week

Proof of the existence of the phase transition via renormalization group transformation. Relevant, irrelevant and marginal parameters. Exemplification of repulsive, attractive and critical fixpoints. Study of the renormalization group equations around a given fixpoint, the deduction technique of critical exponents.

# **Requirements:**

- for a signature

Attendance at **lectures** is recommended, but not compulsory.

Participation at **practice classes** is compulsory. A student must attend the practice classes and may not miss more than three times during the semester. In case a student does so, the subject will not be signed and the student must repeat the course. During the semester there will be two written examinations (on 6th and 13th week) from lectures and practice as well. The requirement for a signature is a successful (> 20%) completion of each written exam.

# - for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have a practical signature.

The grade for the examination is given according to the following table:

Score	Grade	
0-49	fail (1)	
50-62	pass (2)	
63-75	satisfactory (3)	
76-88	good (4)	
89-100	excellent (5)	
Person responsible for course: Dr. Zsolt Gulácsi, professor, DSc		
Lecturer: Dr. Zsolt Gulácsi, professor, DSc		

<b>Title of course</b> : Nanodiffusion and segregation <b>Code</b> : TTFME0141	ECTS Credit points: 3
Type of teaching, contact hours	
- lecture: 2 hours/week	
- practice: 0 hours/week	
- laboratory: -	
Evaluation: exam	
Workload (estimated), divided into contact hours:	
- lecture: 28 hours	
- practice: 0 hours	
- laboratory: -	
- home assignment: 40 hours	
- preparation for the exam: 40 hours	
Total: 108 hours	
Year, semester: 2 <sup>nd</sup> year, 1 <sup>st</sup> semester	
Its prerequisite(s): TTFME0105	
Further courses built on it: TTFME0142	

#### **Topics of course**

Introduction. Continuum models: Classical diffusion theories: Fick's I and II equations; analytical solutions of the II equation in simple cases, assuming an independent diffusion coefficient; Boltzmann transformation, parabolic (square root) (scale) law. Drift, external driving forces (Nernst-Einstein equation). Mutual diffusion. Diffusion in multilayers: The work of DuMond and Youtz; the effect of the concentration dependent diffusion coefficient on the development of the concentration profile (diffusion asymmetry, asymmetric concentration profile, sharpening); high concentration gradients (Cahn-Hilliard theory); stress and diffusion (Stephenson model). Atomic models: Diffusion mechanisms: (vacancy, interstitial, direct exchange, ...); Deterministic kinetic description of diffusion: comparison of continuum and atomistic models (diffusion coefficient - jump frequency); atomic meaning of concentration dependence of diffusion coefficient (diffusion asymmetry); the limitations of the continuum description on the nanoscale, the narrowing of the validity limit of the continuum description with increasing diffusion asymmetry; the effect of chemistry on the behavior of binary alloys (complete mixing, phase separation, ordering, solid state reaction). Anomalous kinetics, nonstoichiometric phases, dissolution of ordered phase. Modeling of diffusion with kinetic Monte Carlo method: differences in deterministic and stochastic descriptions, jump probability; the comparison of the deterministic and kinetic Monte Carlo methods. Segregation: the phenomenon of segregation; surface tension, chemistry and size effect, driving forces of segregation; equilibrium and kinetic segregation isotherms (Henry, McLean, Fowler-Guggenheim); surface segregation in deterministic kinetic and kinetic Monte Carlo models, dissolution of thin layers into substrates. Experimental techniques on the nanoscale: X-ray, synchrotron, neutron-based techniques; surface analytical methods, eg. Auger Electron Spectroscopy (AES), X-ray Photoelectron Spectroscopy (XPS), Atomic Probe Microscopy (APM: AFM, STM)

# Literature

C.Kittel: Introduction to Solid State Physics

J.Philibert:Atom Movements: Diffusion And Mass Transport In Solids (Monographies De Physique) Beke DL, Cserháti Cs, Erdélyi Z, Szabó IA, Segregation in nanostructures (Chapter 7) in Nalwa HS (ed.) Nanoclusters and Nanocrystals - Stevenson Ranch: American Scientific Publishers pp. 211-252 (2003) Schedule: 1<sup>st</sup> week Information, introduction. 2<sup>nd</sup> week Continuum models Analytical solutions of Fick's second equation in simple cases, assuming an independent diffusion coefficient; Boltzmann transformation, parabolic (square root) (scale) law. Drift, external driving forces (Nernst-Einstein equation)  $3^{rd}$  week Continuity models Interdiffusion. Diffusion in multilayers: Beginning with the work of DuMond and Youtz; 4<sup>th</sup> week Continuity models The effect of the concentration dependent diffusion coefficient on the development of the concentration profile (diffusion asymmetry, asymmetric concentration profile, sharpening); 5<sup>th</sup> week Continuity models Interdiffusion in high concentration gradients, Cahn-Hilliard theory; 6<sup>th</sup> week Continuity models Stress and diffusion, Stephenson model 7<sup>th</sup> week Atomic models Diffusion mechanisms: (vacancy, interstitial, direct exchange, ...); Deterministic kinetic description of diffusion: comparison of continuum and atomistic models (diffusion coefficient - jump frequency); the atomic meaning of the concentration dependence of the diffusion coefficient (diffusion asymmetry). 8<sup>th</sup> week Atomic models The limitations of the continuum description on the nanoscale, the narrowing of the validity limit of the continuum description with increasing diffusion asymmetry; the effect of chemistry on the behavior of binary alloys (complete mixing, phase separation, ordering, solid state reaction). Anomalous kinetics, non-stoichiometric phases, dissolution of ordered phases. 9th week Atomic models Modeling fs diffusion with kinetic Monte Carlo method: Differences in deterministic and stochastic descriptions, jump probability; the comparison of deterministic and kinetic Monte Carlo methods. 10<sup>th</sup> week Segregation: The phenomenon of segregation; surface tension, chemistry and size effect as driving forces of segregation motors; equilibrium and kinetic segregation isotherms (Henry, McLean, Fowler-Guggenheim). 11<sup>th</sup> week Segregation: Surface segregation in the deterministic kinetic and kinetic Monte Carlo models, dissolution thin layers into substrates. 12<sup>th</sup> week Experimental techniques on the nanoscale Diffusion: X-ray, synchrotron, neutron-based techniques; 13<sup>th</sup> week

Test methods on the nanoscale Segregation: surface analytical techniques, eg. Auger Electron Spectroscopy (AES), X-ray Photoelectron Spectroscopy (XPS), Atomic Probe Microscopy (APM: AFM, STM) 14<sup>th</sup> week Summary, consultation **Requirements:** - for a signature Attendance at lectures is recommended, but not compulsory. - for a grade The course ends in an **examination**. The requirement for applying for an exam is to have a practical signature. The grade for the examination is given according to the following table: Score Grade 0-49 fail (1) 50-62 pass (2) 63-75 satisfactory (3) 76-88 good(4)89-100 excellent (5)

Person responsible for course: Prof. Dr. Zoltán Erdélyi, full professor, DSc

Lecturer: Prof. Dr. Zoltán Erdélyi, full professor, DSc

Title of course: Magnetism and nanomagnetism Code:	<b>ECTS Credit points:</b> 5	
Type of teaching, contact hours		
- lecture: 2 hours/week		
- practice: 1 hours/week		
- laboratory: 1 hours/week		
Evaluation: practical grade (practice and laboratory), exam		
Workload (estimated), divided into contact hours:		
- lecture: 28 hours		
- practice: 14 hours		
- laboratory: 14		
- home assignment: 14 hours		
- preparation for the exam: 40 hours		
Total: 110 hours		
<b>Year, semester</b> : 1 <sup>st</sup> year, 1 <sup>st</sup> semester		
Its prerequisite(s): - elektromagnetism		
Further courses built on it: -		

# **Topics of course**

Interactions between materials and magnetic field, the most important parameters of magnetic materials, classification and characterization of magnetic materials. Magnetic moments of atoms, dia, para and ferromagnetism. Measurement methods of magnetic parameters, magnetization curves, magnetic hysteresis, phenomenological model of ferromagnetism, Curie-Weiss law. Exchange interaction, exchange energy, exchange integral, Ising-model. Anisotropies; magnetocrystalline anisotropy, shape anisotropy. Ferromagnetic domains, domain walls; Bloch-wall, Neel-wall, reversible and irreversible motion of domain walls, origin of Barkhausen-noise, detection, analysis and applications of Barkhausen-noise. Experimental methods for observation of ferromagnetic domain structures. Soft magnetic materials; crystalline, amorphous, nanocrystalline metallic materials and ferrites. Properties, fabrication and applications of soft ferromagnets. Hard magnetic materials; traditional alloys, rare-earth magnets, hard ferrites, properties, fabrication and applications. Isolated ferromagnetic nanoparticles, superparamagnetism. Spin-glasses, cluster-glasses, magnetic properties of nanocrystalline materials. Related phenomena; magnetostriction, magnetostrictive materials, magnetic shape memory and applications, magnetic resistance, giant and colossal magnetic resistance, applications

Laboratory practices:

Lab.1. Measurement of ferromagnetic hysteresis loops by induction method.

Lab.2. Measurement of magnetization curves by vibration magnetometer.

Lab.3. Measurement of magnetic field by fluxgate magnetometer

Lab.5. Barkhausen noise and magnetic emission measurements, statistical noise analysis.

Lab.6. Industrial Barkhausen noise measurement, determination of stress fields in steels.

Lab.7. Measurement of magnetostriction and magnetic shape memory effect.

#### Literature

# Slides, tests etc. : moodle.phys.unideb.hu\ magnetism and nanomagnetism

# Compulsory:

- C. Kittel: Introduction to Solid State Physics
- K. Kreher: Solid State Physics

# Recommended:

- J. Crangle: Solid State Magnetism
- F. Fiorillo:Measurement and Characterisation of Magnetic Materials
- E. Della Torre: Magnetic Hysteresis
- G. Bertotty: Hysteresis in magnetism
- S. Tumansky: Handbook of Magnetic Measurements

# Schedule:

1<sup>st</sup> week

Materials in magnetic field, properties, classification and characterisation of magnetic materials.

 $2^{nd}$  week

Magnetic moment of an atom

.3<sup>rd</sup> week

Dia, para and ferromagnetism

 $4^{th}$  week

Basics of ferromagnetism, magnetization curve, measurement methods of magnetization curve  $5^{th}$  week

Phenomenological theory of ferromagnetism, Curie-Weiss law

6<sup>th</sup> week

Exchange interaction, Ising-model

7<sup>th</sup> week

Anisotropies in ferromagnetic materials, magnetic domains

 $8^{th}$  week

Experimental methods for observation of domain structures, origin of Barkhausen-noise, measurement methods and applications of Barkhausen noise.

9<sup>th</sup> week

Soft magnetic materials, properties and applications

10<sup>th</sup> week

Hard magnetic materials, properties and applications

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11<sup>th</sup> week
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Ferromagnetic nanoparticles (superparamagnetism)

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12<sup>th</sup> week
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Spin glasses, cluster glasses, amorphous and nanocrystalline materials

13<sup>th</sup> week

Related phenomena, magnetostriction, magnetic shape memory, magnetic resistance, giant magnetic resistance (GMR)

14<sup>th</sup> week

Summary, discussion

Laboratory practices in small groups by individual timetable.

# **Requirements:**

- for a signature

Attendance at **lectures** is recommended, but not compulsory.

Participation at **practice classes** is compulsory. A student must attend the practice classes and may not miss more than three times during the semester. In case a student does so, the subject will not be signed and the student must repeat the course.

During the semester there are practical home works which have to be submitted. The requirement for a signature is a successful (> 50%) completion of the home works.

#### - for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have practical signatures and grades at least 2 (practice and laboratory practice as well).

The grade for the examination is given according to the following table:

Score	Grade
0-49	fail (1)
50-62	pass (2)
63-75	satisfactory (3)
76-88	good (4)
89-100	excellent (5)

Person responsible for course: Dr. Lajos Daróczi, associate professor, PhD

Lecturer: Dr. László Tóth, assistant professor, PhD

<b>Title of course</b> : Measurement of materials properties <b>Code</b> : TTFML0144	ECTS Credit points: 2	
Type of teaching, contact hours		
- lecture: 0 hours/week		
- practice: 0 hours/week		
- laboratory: 2 hours/week		
* <b>Evaluation:</b> signature + grade for lab.report		
Workload (estimated), divided into contact hours:		
- lecture:		
- practice:		
- laboratory: 32		
- home assignment: 32 hours		
- preparation for the exam:		
Total: 64 hours		
Year, semester: Condensed Matter course block		
Its prerequisite(s): Condensed Matter 3. TTFME0105		
Further courses built on it: -		
Topics of course		
to expand the knowledge about the subject with the help of 8 4-hour measurement exercises taken from the topic of condensed materials.		
Literature		
Compulsory: The students are provided by 10-20 page measurement instructions prepared by the Institute. Recommended:		
Schedule:		
1 <sup>st</sup> week		
Information, introduction, accident and safety instruction, discussion $2^{nd}$ week	of laboratory schedule.	
2 week Metallographic studies by light microscopy.		
3 <sup>rd</sup> week		
Examination of surface and composition by scanning electron microscopy. $\mathcal{A}^{th}$ weak		
Structural examination by transmission electron microscopy		
5 <sup>th</sup> week		
Preparation of thin films and depth analysis of the prepared specimens by secondary neutral mass spectrometry		
Structural analysis by X-ray diffraction		
7 <sup>th</sup> week		
7 week Investigation of ferromagnetic materials using Barkhausen noise method		
8 <sup>th</sup> week		

Preparation of alloys by arc melting method 9<sup>th</sup> week

Surface studies with SPM and AFM equipment.

# **Requirements:**

\* • the basic knowledge of the laboratory practice theory, the measurement, the preparation of a measurement report in electronic form: sufficient;

• accurate knowledge of the theory of exercises, carrying out the measurement, making a measurement report in electronic form: medium;

• Basic knowledge of laboratory practice theory, accurate measurement and evaluation of measurements, preparation of measurement report in electronic form: good;

• accurate knowledge of laboratory practice theory, accurate measurement and evaluation of measurements, preparation of measurement report in electronic form: excellent.

Person responsible for course: Dr. Csaba Cserháti, associate professor, PhD

Lecturer: Dr Csaba Cserháti and Dr. Lajos Daróczi, associate professor, PhD
<b>Title of course</b> : Transmission and analytical electronmicroscopy <b>Code</b> : TTFME0146	ECTS Credit points: 5
Type of teaching, contact hours	
- lecture: 2 hours/week	
- practice: 1 hours/week	
- laboratory: 1 hours/week	
* Evaluation: exam	
Workload (estimated), divided into contact hours:	
- lecture: 28	
- practice: 14	
- laboratory: 14	
- home assignment: 14 hours	
- preparation for the exam: 40	
Total: 100 hours	
Year, semester: Condensed Matter course block	
Its prerequisite(s):	
Further courses built on it: -	
Topics of course	
To get acquainted with the most important method used in transmission electron microscopy used in modern materials science, with special regard to structural analysis	
Literature	
Compulsory: Williams, David B., Carter, C. Barry: Transmiss Textbook for Materials Science Recommended:	ion Electron Microscopy A
Schedule:	
1 <sup>st</sup> week	
Information, introduction. The place of microscopy in materials science $2^{nd}$ week	e
Presentation of the principles and operating methods of atomic or near-a microscopes (SPM, AFM, etc.) $3^{rd}$ week	tomic resolution scanning probe
Structure and operation modes of the transmission electron microscope $4^{th}$ week	
Repetition and extension of the basic crystallological concepts necessar diffraction and the reciprocal lattice space.	ry for the discussion of electron
S <sup>***</sup> week	
Kinematic theory of electron diffraction and its application 1.	
Kinematic theory of electron diffraction and its application 2. $7^{th}$ week	
Dynamatic theory of electron diffraction and its application 1.	

 $\delta^{th}$  week Dynamatic theory of electron diffraction and its application 2.  $9^{th}$  week Electron diffraction phenomena, Kikuchi lines  $10^{th}$  week Convergent beam electron diffraction  $11^{th}$  week Theory of image formation  $12^{th}$  week X-ray microanalysis in the TEM  $13^{th}$  week Electron Energy Loss Spectroscopy (EELS) in the TEM  $14^{th}$  week Summary, consultation

# **Requirements:**

\*

\* • Participation in practical and laboratory activities, the condition for admission to the exam is the successful completion of the related practice and laboratory

\* • Knowledge of the laws, items and definitions related to the topic in the exam: sufficient;

\* • in addition, knowledge of the main steps in deriving the most important theories and regularities: medium;

\* • in addition, deriving the derivations with less help, seeing the connections: good;

• in addition, derivation of derivations without help, ability to apply them: remarkable.

Person responsible for course: Dr. Csaba Cserháti, associate professor, PhD

Lecturer: Dr. Lajos Daróczi, associate professor, PhD

<b>Title of course</b> : Environmental Physics 3. <b>Code</b> : TTFME0153	ECTS Credit points: 4
Type of teaching, contact hours - lecture: 2 hours/week	
- practice: 1 hours/week	
- laboratory: -	
<b>Evaluation:</b> written or oral exam (and Implementation of a project tasl	k)
Workload (estimated), divided into contact hours:	
- lecture: 28 hours	
- practice: 14 hours	
- laboratory: -	
- home assignment: 40 hours	
- preparation for the exam: 40 hours	
Total: 122 hours	
Year, semester: 2 <sup>nd</sup> year, 1 <sup>st</sup> semester	
Its prerequisite(s): -	
Further courses built on it: -	
Topics of course	
Environment and risk, Cosmic radiation environment, Radon in the environment, Environmental effects of nuclear reactor accidents, Climate change, Environmental noise, Volcanoes of the Carpathian-Pannonian region, Aerosol pollution in the atmosphere, Extreme aerosol effects, Normal emissions of nuclear power plants; Disposal of radioactive waste, Flows in the environment, water resources and threats, Water infiltration, age determination, Alternative energy sources, Stabil isotope ratio measurements: $\delta^{13}$ C, $\delta^{18}$ O, $\delta^{15}$ N, $\delta^{34}$ S, $\delta$ D	
Literature	
Compulsory:	
Text distributed in Moodle environment	
Schedule: 1 <sup>st</sup> week Environment and risk	
2 <sup>nd</sup> week	
2 <sup>rm</sup> week	
Cosmic radiation environment	
3 <sup>rd</sup> Week	
Lab: Etchod track detector measurements	
A <sup>th</sup> week	
Finite Free and the second sec	
Lab: Analytics of environmental radioisotopes (gamma spectrometry)	
Las. Thaytes of environmental factorsotopes (gamma spectrometry)	
5 <sup>th</sup> week	
Climate change	

6 <sup>th</sup> week
Environmental noise
7 <sup>th</sup> week
Volcanoes of the Carpathian-Pannonian region
$\delta^{th}$ week
Aerosol pollution in the atmosphere
9 <sup>th</sup> week
Extreme aerosol effects
Lab: PIXE
10 <sup>th</sup> week
Normal emissions of nuclear power plants; Disposal of radioactive waste
Lab: LSC measurement technique
11 <sup>th</sup> week
Flows in the environment, water resources and threats
12 <sup>th</sup> week
Water infiltration, age determination
Lab: Noble gas mass spectrometry
13 <sup>th</sup> week
Alternative energy sources
14 <sup>th</sup> week
Stabil isotope ratio measurements: $\delta^{13}$ C, $\delta^{18}$ O, $\delta^{15}$ N, $\delta^{34}$ S, $\delta$ D
Lab: MICADAS

## **Requirements:**

- for a signature

Attendance at **lectures** is recommended, but not compulsory.

Participation at **practice classes** is compulsory. A student must attend the practice classes and may not miss more than three times during the semester. In case a student does so, the subject will not be signed and the student must repeat the course.

During the semester there are practical home works which have to be evaluated and submitted by the end of the  $14^{th}$  week of the semester. The requirement for a signature is a successful (> 50%) completion of the home works.

- for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have a practical signature.

The grade for the examination is given according to the following table:

Score	Grade
0-49	fail (1)
50-62	pass (2)
63-75	satisfactory (3)
76-88	good (4)
89-100	excellent (5)

Person responsible for course: Dr. Eszter Baradács, assistant professor, PhD

#### Lecturers:

Dr. Eszter Baradács, assistant professor, PhD; Dr. László Palcsu, senior research associate, PhD, Dr. Zsófia Kertész, senior research associate, , PhD; Dr. Futó István, research associate PhD; Dr. Zsolt

Benkó, research associate PhD; Dr. István Csige, senior research associate PhD; Dr. Róbert Janovics, research associate, PhD;

<b>Title of course</b> : Simulation of Environmental Processes <b>Code</b> : TTFME0154	ECTS Credit points: 4
Type of teaching, contact hours	
- lecture: 2 hours/week	
- practice: 1 hours/week	
- laboratory: -	
Evaluation: written or oral exam (and Implementation of a project task	x)
Workload (estimated), divided into contact hours:	
- lecture: 28 hours	
- practice: 14 hours	
- laboratory: -	
- home assignment: 40 hours	
- preparation for the exam: 40 hours	
Total: 122 hours	
<b>Year, semester</b> : 2 <sup>nd</sup> year, 1 <sup>st</sup> semester	
Its prerequisite(s): -	
Further courses built on it: -	
Topics of course	
The goal and role of computer simulation in environmental science limitations of computer simulations in problem solving in environmer model selection. Reliability of modelling, international intercomp parameter sensitivity and parameter uncertainty. Compartment models.	es. Advantages, constrains and ntal sciences. Modelling types, arison exercises. Analysis of
Behaviour of radionuclides in the environment. Modelling transport of radionuclides in the atmosphere. Subsurface transport. Contamination of the food chain. Groundwater flow and contaminant transport	
Literature	
Compulsory:	
Text distributed in Moodle environment	
<i>Recommended:</i> Schnoor JL (1996) Environmental modeling: fate and transport of po Wiley, New York Nirmalakhandan N (2002) Modeling tools for environmental engine Raton, FL	ollutants in water, air, and soil.
Schedule:	
$1^{st}$ week Introduction. The goal and role of computer simulation in environment limitations of computer simulation. $2^{nd}$ week	al sciences. Constrains and
Introduction to Sail ab: Sail ab basics satur of Sail ab Sail ab as an a	dyancad calculator graphing in

Introduction to SciLab: SciLab basics, setup of SciLab, SciLab as an advanced calculator, graphing in 2D.

 $3^{rd}$  week

Basics of SciLab programming: SciNotes, Definition and use of user functions, cycles, export of data to files

4<sup>th</sup> week

Behaviour of radionuclides in the environment. The transport equation. 5<sup>th</sup> week Modelling transport of radionuclides in the atmosphere 6<sup>th</sup> week Modelling transport processes in soil 7<sup>th</sup> week Contamination of the food chain 8<sup>th</sup> week Pollution of the water environment. Basic equations of water flow and contaminant transport. Theoretical background of modelling. 9<sup>th</sup> week Differential equation in SciLab 1. 10<sup>th</sup> week Differential equation in SciLab2. 11<sup>th</sup> week Differential equation in SciLab 3-4. 12<sup>th</sup> week Implementation of a project task 13<sup>th</sup> week Implementation of a project task 14<sup>th</sup> week Atmospheres of the planets and of some moons of the Solar System. **Requirements:** - for a signature Attendance at **lectures** is recommended, but not compulsory. Participation at practice classes is compulsory. A student must attend the practice classes and may not

miss more than three times during the semester. In case a student does so, the subject will not be signed and the student must repeat the course.

During the semester there are practical home works which have to be evaluated and submitted by the end of the  $14^{th}$  week of the semester. The requirement for a signature is a successful (> 50%) completion of the home works.

- for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have a practical signature.

The grade for the examination is given according to the following table:

Score	Grade
0-49	fail (1)
50-62	pass (2)
63-75	satisfactory (3)
76-88	good (4)
89-100	excellent (5)

# Person responsible for course: Dr. Eszter Baradács, assistant professor, PhD

#### **Lecturers:**

Dr. Eszter Baradács, assistant professor, PhD

<b>Title of course</b> : Radiation protection and dosimetry <b>Code</b> : TTFME0151	ECTS Credit points: 4
Type of teaching, contact hours	
- lecture: 2 hours/week	
- practice: -	
- laboratory: 1 hours/week	
Evaluation: exam	
Workload (estimated), divided into contact hours:	
- lecture: 28 hours	
- practice: -	
- laboratory: 16 (4x4) hours	
- home assignment: 14 hours	
- preparation for the exam: 40 hours	
Total: 98 hours	
<b>Year, semester</b> : 1 <sup>st</sup> year, 2 <sup>nd</sup> semester	
Its prerequisite(s): -	
Further courses built on it: -	
Topics of course	
Overview of measuring instruments and methods for measuring radioactivity, ionizing radiation and radiation dose. The interaction l	and estimating environmental between ionizing radiation and

radioactivity, ionizing radiation and radiation dose. The interaction between ionizing radiation and matter. The chemical and biological processes generated by the radiation in the living material and their biological effects. Stochastic and deterministic effects. Dosimetry concepts. Exposure to population due to natural and artificial radiation sources. Principles, methods and tools of radiation protection. Radiation protection norms and legal regulation. Dosimetry measurements in laboratory practice (environmental gamma dose rate measurement, radon dose estimation, scaling of shielding against gamma radiation).

#### Literature

Compulsory:

ICRP Publication 103: The 2007 Recommendations of the International Commission on Radiological Protection

IAEA General Safety Requirements Part 3, Radiation Protection and Safety of Radiation Sources: International Basic Safety Standards

Recommended:

RADIATION ONCOLOGY PHYSICS: A HANDBOOK FOR TEACHERS AND STUDENTS, IAEA, VIENNA, 2005

# Schedule:

1<sup>st</sup> week

Ionizing and non-ionizing radiations in the environment. Origin and properties of these radiations. Primary cosmic rays and showers of secondary particles generated by the primary ones in the Earth's atmosphere. The origin of the radionuclides present in the environment, their quantitative environmental distribution and the nuclear radiations arising from their radioactive decay. External and internal radiations (from the point of view of the human body).

 $2^{nd}$  week

The interaction between radiation and matter. Interaction of the ionizing radiation with the material. Special features of the interaction of alpha, beta, gamma, neutron and X-ray radiation. Changes in the material: ionization, excitation, ionization density, breakdown of chemical bonds, recombination of ions, de-excitation of atoms, growth of internal energy, radiation damage. Changes in radiation characteristics: loss of energy, scattering, absorption, range. The dependence of the former on the type of radiation and the material quality.

# 3<sup>rd</sup> week

The concept and types of radiation exposure. Natural, occupational, medical, accidental, potential exposure. Radiation dose, as a measure of the extent of lesions occurring in the ex-posed material. Different dose quantities and their units of measure, the considerations that led to their creation. Kerma, absorbed dose, equivalent dose, effective dose, collective dose. The role of radiation type, the weight factor of radiation. The role of radiation absorbing material (live or inanimate, type of living body tissue), the weight factor of body tissues or organs. The role of time, dose rate. Organ dose and whole body effective dose. The role of the location of the radiation source: external or internal dose. Committed dose.

4<sup>th</sup> week

Possibilities and limitations of experimental determination of dose. Physical basics of the operation of the experimental tools and methods used to measure dose. Operation of radiation detectors applied to dose measurements. Ionization chamber, proportional counter, Geiger-Müller counter. The proportionality of the counting rate with dose rate. Dosimeters based on thermo-luminescence, photochemical transformation and particle track detection. Proportionality of the measure of luminescence, dimming and track density with dose. Personal dosimeters: film dosimeter, thermos-luminescent dosimeter, pen dosimeter, solid state track detector. Problems with the calibration of dosimeters. Determination of internal dose based on measurements using whole body counters.

# $5^{th}$ week

Possibilities and limitations of the determination of dose by calculations. Estimation of external dose by calculations. How the amount and internal distribution of the external dose de-pend on the type of radiation? The spatial distribution of dose near point sources of beta and gamma radiation. Production of simple spatial dose distributions for calibration of dosimeters, estimation of corrections by calculations. The spatial distribution of dose from a radiation source that is distributed evenly in half-space. The spatial distribution of dose from a radiation source that is distributed evenly over an endless plane. The assumptions and calculations required to estimate internal dose. Estimate of dose from inhalation (radiation source deposited on the inner surface of the airways). Estimation of dose from ingestion.

# $6^{th}$ week

The amount of annual human dose (obtained from ionizing radiation) from different environmental sources. Doses from natural sources. Dose from cosmic rays. Doses from the radiation of radionuclides decaying outside the body and within the body. Doses arising from dietary intake and inhaled radionuclides. World average doses and regional differences. Doses from artificial sources. Doses from medical applications of ionizing radiation and radioisotopes. Doses from artificial radioactivity released into the free environment. War and accident doses. Occupational doses.

# $7^{th}$ week

Direct microbiological effects of ionizing radiations. Formation of free radicals. Damage to biomolecules: DNA breaks, membrane injuries. Correcting DNA damage. Cell death. Changes in cell cycle regulation. Physical, chemical and biological factors that modify the microbiological effects of radiation (radiation type, dose rate, temperature, oxygen effect, water content, amount of antioxidants, cell cycle state, age, gender). Radiation sensitivity of the living world.

# $8^{th}$ week

Physiological changes, symptoms, illnesses and death caused by ionizing radiations depending on the dose. Increase in the frequency of malignant tumors and mutations in the population due to a moderate extra dose. Randomness and risk of effects. Molecular and cell biology mechanisms of malignant tumor

formation. Biological effects of low doses: doubts about the linearity between the dose and the triggered effect (risk).

# 9<sup>th</sup> week

Certainly and within a short period of time occurring symptoms and bodily changes in case of large extra doses. Diseases of some organs and their dose thresholds (immune system, gastro-intestinal tract, genitals, skin, vascular system, central nervous system, lens of the eye). Symptoms and clinical course of acute radiation disease. Median lethal and lethal dose. Late residual effects. Local radiation injuries. Radiation accidents involving personal injury and the lessons learned from these cases.

# $10^{th}$ week

Environmental quantities, biological effects and dosimetry of non-ionizing electromagnetic radiations. Power density of ultraviolet, visible and infrared radiation from the Sun. The power density of artificial electromagnetic radiations depending on the distance from various electrical equipment. Dosimetry quantities: specific (per mass unit or surface unit) absorbed energy and power. Methods for measuring dosimetric quantities. Absorption ability of human body material as a function of frequency. Free radicals, molecular breakdowns, generation of electricity and heat in the body. Physiological changes, symptoms.

# $11^{th}$ week

The concept of radiation protection, regulation of radiation-related activities at a social level to protect human health. Basic principles and basic concepts of radiation protection. Requirement of justification for radiation hazardous activity, principle and practice of optimizing the protection against radiation, application of individual dose limits, exemption. Deriving dose limits from the estimated risks. Intervention to reduce radiation exposure. Radiation protection for radiological sources. Radiation protection of the population. Exemption and intervention levels and dose limits for workers with radiation sources and population.

# $12^{th}$ week

Practical methods of protection against ionizing radiation and reducing radiation exposure. Precautionary measures to avoid the occurrence of radiation exposure. Radiation hazardous occupations. Guidelines for the design of radiation hazardous workplaces, work safety regulations. Personal dosimetry. Methods for protection against external radiation. Time protection, distance protection, radiation absorbing protective layers. Shielding alpha, beta, gamma and neutron radiation. Protection against internal radiation. Rules of work with open radioactive preparations. Radiation protection inspections and radiation protection education.

# $13^{th}$ week

Possibilities for reducing exposure to the population. Methods to reduce the radon concentration in residential areas. Possibilities and ways of reducing the population exposure to radiations applied in medical interventions. Ways of intervention to protect the population in the event of the use of a nuclear weapon or a nuclear installation accident. Isolation, rescue, iodine prophylaxis, migration, restriction of food consumption, radiological decontamination. Radiation protection control of the natural environment. National control programs. Checking the environment of nuclear facilities. Experience with nuclear accidents and radiation source accidents.

# 14<sup>th</sup> week

International and national organizational system of radiation protection. Recommendations from the International Commission on Radiological Protection (ICRP), the role of the International Atomic Energy Agency (IAEA) and other international organizations. The EURATOM Directive. The institutional system of radiation protection in Hungary. The legal regulation of radiation protection activity. The 1996 CXVI. Act on Nuclear Energy and the government decrees on its implementation.

## **Requirements:**

#### - for a signature

Attendance at **lectures** is recommended, but not compulsory.

Participation at **laboratory sessions** is compulsory. The student must attend all the four sessions. In case the student doesn't so, the course will not be signed and the student must repeat it. Attendance at laboratory sessions will be recorded by the session leader. Being late is equivalent with an absence. Students are required to bring drawing instruments to each sessions. Active participation is evaluated by the teacher. If a student's behavior or conduct doesn't meet the requirements of active participation, the teacher may evaluate his/her participation as an absence.

The student will obtain partial grades for all the four laboratory sessions one by one. The partial grades go from fail (1) to excellent (5) at the discretion of the session leader. For the laboratory signature, the student has to complete all the laboratory sessions with a partial grade better than fail (1). If the latter condition is not met then the student must repeat the laboratory sessions.

#### - for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have a laboratory signature.

The grade for the examination is given according to the following table:

Score	Grade
0-49	fail (1)
50-62	pass (2)
63-75	satisfactory (3)
76-88	good (4)
89-100	excellent (5)

Person responsible for course: Dr. Zoltán Papp, associate professor, PhD

Lecturer: Dr. István Csige, associate professor, PhD

<b>Title of course</b> : Physics of the Atmosphere <b>Code</b> : TTFME0155	ECTS Credit points: 3
Type of teaching, contact hours	
- lecture: 2 hours/week	
- practice: -	
Evaluation: exam	
Workload (estimated), divided into contact hours:	
- lecture: 28 hours	
- practice: -	
- laboratory: -	
- home assignment: 40 hours	
- preparation for the exam: 40 hours	
Total: 108 hours	
Year, semester: 1 <sup>st</sup> year, 2 <sup>nd</sup> semester	
Its prerequisite(s): -	
Further courses built on it: -	
Topics of course	
The formation and evolution of the atmosphere. Atmospheres of the planets of the Solar System. Composition of Earth's atmosphere (permanent and variable components; origin, point and extended sources; geogases; biogenic and anthropogenic sources). Vertical and horizontal motions in the atmosphere and their effects on the physical-chemical properties of the atmosphere (thermal convection, advection, diffusion). Application of some cosmogenic radioactive isotopes (Be-7, T) in the study of motion and mixing of air masses. Calculation of atmospheric trajectories and of spatial and temporal variations of concentrations.	
Literature	
<i>Compulsory:</i> John Houghton: The physics of atmospheres. Cambridge University Pr <i>Recommended:</i> The Atmosphere. Ed.: R. F. Keeling, Elsevier, 2006.	ess
Schedule:	
1 <sup>st</sup> week	
The formation and evolution of the atmosphere.	
2 <sup>nd</sup> week	
Composition of Earth's atmosphere (permanent and variable compone sources; geogases; biogenic and anthropogenic sources). $3^{rd}$ week	ents; origin, point and extended
Thermodynamics. Barometric formula of the atmospheric pressure. Dr adiabatic lapse rate. $4^{th}$ week	y adiabatic lapse rate and moist

Vertical and horizontal motions in the atmosphere and their effects on the physical-chemical properties of the atmosphere (thermal convection, advection, diffusion).

 $5^{th}$  week

Radiative transfer in the atmosphere

 $6^{th}$  week

Sampling of aerosol. Experimental methods to determine mass concentration, size distribution and chemical composition. Optical properties of aerosol (scattering and absorption of light on aerosol).  $7^{th}$  week

Sources and spatial and temporal variations of aerosol, range of concentrations, environmental and health effects.

 $8^{th}$  week

Application of some cosmogenic radioactive isotopes (Be-7, T) in the study of motion and mixing of air masses. Calculation of atmospheric trajectories and of spatial and temporal variations of concentrations

# 9<sup>th</sup> week

Analytics of carbon dioxide (modern procedures based on sensors). Stable isotope composition of carbon dioxide, carbon and oxygen mass spectrometry.

 $10^{th}$  week

Global cycle of carbon. Isotope effects in the global carbon cycle; flux determination from measurements of micrometeorology and isotope ratios.

11<sup>th</sup> week

Global water cycle. Water in the atmosphere, precipitation, evapotranspiration. Stable isotope composition of water, analytical mass spectrometry.

 $12^{th}$  week

Physics of isotopes of water: fractionation processes and their application in paleoclimatology.

 $13^{th}$  week

Physical analytics of carbon monoxide and noble gases. Application of tracer gases in the study of transport processes in the atmosphere.

14<sup>th</sup> week

Atmospheres of the planets and of some moons of the Solar System.

# **Requirements:**

- for a signature

Attendance at **lectures** is recommended, but not compulsory.

During the semester there are practical home works which have to be evaluated and submitted by the end of the  $14^{th}$  week of the semester. The requirement for a signature is a successful (> 50%) completion of the home works.

- for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have a practical signature.

The grade for the examination is given according to the following table:

Score	Grade
0-49	fail (1)
50-62	pass (2)
63-75	satisfactory (3)
76-88	good (4)
89-100	excellent (5)

# Person responsible for course: Dr. István Csige, associate professor, PhD

# Lecturers:

Dr. István Csige, associate professor, PhD Dr. Elemér László, associate professor, PhD

<b>Title of course</b> : Measurements in environmental physics <b>Code</b> : TTFML0156	ECTS Credit points: 2
Type of teaching, contact hours	
- lecture: -	
- practice: -	
- laboratory: 2 hours/week	
Evaluation: grade for written laboratory record	
Workload (estimated), divided into contact hours:	
- lecture: -	
- practice: -	
- laboratory: 28 (7x4) hours	
- home assignment: 14 hours	
- preparation for the exam: -	
Total: 42 hours	
<b>Year, semester</b> : 1 <sup>st</sup> year, 2 <sup>nd</sup> semester	
Its prerequisite(s): -	
Further courses built on it: -	
Topics of course	
Study of photovoltaic phenomena. Experiments with microwaves. S handheld GPS device. Measurement of ambient noise. Measurement of Measurement and analysis of weather variables.	Solar panels. Learn and use a f the density of small particles.
Each of the above measurements has duration of 4 hours and the laborat 4-hour blocks in one half of the semester.	tory practice is therefore kept in
Literature	
<i>Compulsory:</i> Mandatory laboratory syllabuses for each practice required by the labo <i>Recommended:</i> For each practice, the data sources recommended by the laboratory ses	ratory sessions leader. sions leader.
Schedule:	

The themes cannot be recorded weekly in advance for all of the student because each student does their own practice alone, the same week each student performs different exercise and the order of the exercises will be different for each student. At the beginning of the semester, a schedule will be made as to which week which student will do which exercise.

# 1<sup>st</sup> exercise

Measurement of photovoltaic phenomena. Instruments used: mercury vapor discharge tube; semiconductor photodiode; optical devices; voltage amplifier; voltmeter. Students are familiar with the photovoltaic phenomenon, with the experimental tools required to study it and experimentally study the dependence of electric voltage on the photovoltaic phenomenon from the wavelength of the light.

# $2^{nd}$ exercise

Studying the properties of microwaves. Tools used: klystron with waveguide tube, horn antenna and power supply (transmitter); dipole antenna with horn antenna (receiver); amplifier and sound generator loudspeaker; optical devices. The student will become familiar with the production of electromagnetic

microwaves and their main properties. Studies the microwaves reflection on a metal plate, refraction at the air/paraffin boundary, polarization with grid, diffraction by an obstacle. The student determines the wavelength of the microwaves in a Michelson interferometer arrangement and studies the waveguide ability of the waveguide tube.

# $3^{rd}$ exercise

Investigating the operation of solar cells. Tools used: solar cells; incandescent lamp; photo-resistor; multimeters; potentiometer; DC power supply. The student becomes acquainted with the operation of the solar cell and the accompanying devices suitable for studying it. The student is experimentally studying the dependence of the conductivity of the photo-resistor on the intensity of the illumination and the characteristics of the solar cell (how the idle voltage and the short-circuit current depend on the intensity of the illumination).

# 4<sup>th</sup> exercise

Learn and use a handheld GPS device. Tools used: handheld GPS device; data processing computer with suitable software; printer. The student becomes acquainted with the principle, implementation and application of the global satellite positioning method and the operation of the device he uses. The student gets acquainted with the need for unobstructed access to the satellites, with the most important settings of the device. Executes waypoints along a route on a map, try the tracking function. The student records a route and navigates through a navigational aid with a pre-recorded route, measures altitude, searches for a previously recorded location, and stores, records, and prints data recorded by the computer.

# 5<sup>th</sup> exercise

Measurement of ambient noise. Devices used: handheld noise measuring device; calibration device; computer with suitable software. The student stands out with the concept of noise, the principle of noise measurement, its execution, and the operation of the instrument he uses. The student calibrates the device, then examines the urban traffic noise. In this context the student measures the maximum noise level or makes longer noise recordings at different public areas. The on-site noise recordings are subsequently evaluated in the laboratory by computer soft-ware. Draws conclusions from the measurement results obtained.

# 6<sup>th</sup> exercise

Measurement of the density of small particles. Tools used: pycnometer; balance; distilled water flask; ultrasonic thermostat; filter paper; rock granulate of unknown density. The student determines the work volume of the pycnometer. The student pour a suitable amount of rock granulate into the pycnometer and measures the mass of the rock granulate with balance. Then fills the pycnometer with distilled water up to a mark while measuring the water volume. Then calculates the bulk density of the rock granulate from the measured mass and volumes. Repeats the measurement and calculation after ultrasonic ventilation and compares the result to the previous one. Draws conclusions from the results obtained. 7<sup>th</sup> exercise

Measurement and analysis of weather variables. Tools used: compact automatic weather station; computer with suitable software. The student becomes acquainted with the basic principles of measurement of weather variables, with the necessary tools and the construction and operation of the measuring system he uses. The student reads from the console of the station and records the current values of the weather variables. From the console's memory, the student uploads data from the last one week to the computer and gets acquainted with the weather station software. Using the software gives a textual description about the weather of his/her recent birthday (in Debrecen), calculates the average temperature of the previous day and then calculates the absolute humidity of the air at some different times. Draws conclusions from the results.

# **Requirements:**

#### - for a signature

Participation at laboratory sessions is compulsory. A student must attend all the seven sessions. In case a student doesn't so, the course will not be signed and the student must repeat it. Attendance at laboratory sessions will be recorded by the session leader. Being late is equivalent with an absence. Students are required to bring drawing instruments to each sessions. Active participation is evaluated by the teacher. If a student's behaviour or conduct doesn't meet the requirements of active participation, the teacher may evaluate his/her participation as an absence.

- for a grade

The student will obtain grades for all the exercises one by one. The grades go from fail (1) to excellent (5) at the discretion of the session leader. The grade of the course will be the arithmetic mean of the grades obtained for each exercises rounded to the full, provided that the student has completed all the exercises with a grade better than fail (1). If the latter condition is not met then the grade of the course is fail (1) and the student must repeat the course in conformity with the EDUCATION AND EXAMINATION RULES AND REGULATIONS.

Person responsible for course: Dr. Zoltán Papp, associate professor, PhD

Responsible instructor: Dr. Eszter Baradács dr. Erdélyiné, assistant professor, PhD

<b>Title of course</b> : Advanced Nuclear Physics <b>Code</b> : TTFME0161	ECTS Credit points: 4
Type of teaching, contact hours - lecture: 2 hours/week - practice: 1 hours/week - laboratory: -	
Evaluation: exam	
Workload (estimated), divided into contact hours: <ul> <li>lecture: 28 hours</li> <li>practice: 14 hours</li> <li>laboratory: -</li> <li>home assignment: 40 hours</li> <li>preparation for the exam: 40 hours</li> <li>Total: 122 hours</li> </ul> Year, semester: 1 <sup>st</sup> year, 2 <sup>st</sup> semester Its prerequisite(s): - Further courses built on it: -	
Topics of course	
Nuclear structure. The basic idea of the shell model, harmonic os symmetry, spin-orbit interaction, magic numbers. Deformed nucle breaking, liquid drop model of vibration and rotation. Elliott mo dynamically breaking symmetry. Wigner's supermultiplets, quartets, generating and dynamic algebras. Boson model of collectivity, sup nuclei, cold quantum phases, quasi-dynamic symmetry. Microscopic between shell, collective and cluster model, multichannel symmetry shell model. Other structure models. Nuclear reactions. Compound Pre-equilibrium model. Nuclear decay, limits of stability. Exotic radi	scillator, geometric and dynamic ar states, spontaneous symmetry del: shell model of collectivity, nuclear cluster model, spectrum ersymmetry. Phase transitions in model of collectivity, relationship . Ab initio method, the truncated nucleus model. Direct reactions. oactivity. Spontaneous fission.

# Literature

Recommended:

H.J. Lipkin: Lie groups for pedestrians, North-Holland Amsterdam, 1966

S.S.M.Wong: Nuclear Physics, John Wiley & Sons NY, 1998

D.J. Rowe, J.L. Wood: Fundamentals of Nuclear Models, World Scientific New Jersey, 2010

T Dytrich, K. D. Sviratcheva, J. P. Draayer, C. Bahri and J. P. Vary: Ab initio symplectic no-core shell model, TOPICAL REVIEW, J. Phys. G. Nucl. Phys. 35 (2008) 123 101

P. Van Isacker: Symmetries in Nuclei, arXiv:1012.3611v1 [nucl-th]

J. Cseh: Algebraic models for shell-like quarteting of nucleons, Physics Letters B 743 (2015) 213

# Schedule:

1<sup>st</sup> week

Nuclear structure. The basic idea of the shell model, harmonic oscillator, geometric and dynamic symmetry.

 $2^{nd}$  week

Spin-orbit interaction, magic numbers.
3 <sup>rd</sup> week
Deformed nuclear states, spontaneous symmetry breaking, liquid drop model of vibration and rotation.
4 <sup>th</sup> week
Elliott model: shell model of collectivity, dynamically breaking symmetry.
5 <sup>th</sup> week
Wigner's supermultiplets, quartets.
$6^{th}week$
Nuclear cluster model, spectrum generating and dynamic algebras.
7 <sup>th</sup> week
Boson model of collectivity, supersymmetry.
$\delta^{th}$ week
Phase transitions in nuclei, cold quantum phases, quasi-dynamic symmetry.
9 <sup>th</sup> week
Microscopic model of collectivity, relationship between shell, collective and cluster model, multichannel symmetry.
10 <sup>th</sup> week
Ab initio method, the truncated shell model. Other structure models.
11 <sup>th</sup> week
Nuclear reactions. Compound nucleus model.
12 <sup>th</sup> week
Direct reactions. Pre-equilibrium model.
13 <sup>th</sup> week
Nuclear decay, limits of stability. Exotic radioactivity. Spontaneous fission.
14 <sup>th</sup> week
Consultation.
Requirements

*Requirements: - for a signature* 

Attendance at **lectures** is recommended, but not compulsory.

Participation at **practice classes** is compulsory. A student must attend the practice classes and may not miss more than three times during the semester. In case a student does so, the subject will not be signed and the student must repeat the course.

During the semester there are practical home works which have to be evaluated and submitted by the end of the  $14^{th}$  week of the semester. The requirement for a signature is a successful (> 50%) completion of the home works.

- for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have a practical signature.

The grade for the examination is given according to the following table:

Score	Grade
0-49	fail (1)
50-62	pass (2)
63-75	satisfactory (3)
76-88	good (4)
89-100	excellent (5)

Person responsible for course: Dr. Judit Darai, associate professor, PhD

Lecturer: Dr. Judit Darai, associate professor, PhD

<b>Title of course</b> : Nuclear technologgy <b>Code</b> : TTFME0162	ECTS Credit points: 4
Type of teaching, contact hours	
- lecture: 2 hours/week	
- practice: 1 hours/week	
- laboratory: -	
Evaluation: exam	
Workload (estimated), divided into contact hours:	
- lecture: 28 hours	
- practice: 14 hours	
- laboratory: -	
- home assignment: 40 hours	
- preparation for the exam: 40 hours	
Total: 122 hours	
Year, semester: 1 <sup>st</sup> year, 2 <sup>nd</sup> semester	
Its prerequisite(s): -	
Further courses built on it: -	
Topics of course	
Characterization of nuclear states and processes. Data measurements in	nuclear physics. Interaction of

radiations with matter: light and heavy charged particles, neutral particles, gamma-radiation. Detection and spectrometry, characteristic parameters of the instruments. Operation basics of radiation detectors: creation of electric charge (gas, semiconductor), light production (scintillation, Cherenkov and transition radiations). Methods of activity measurements in various time domains. Determination of energy, momenta and velocity. Features and devices of gamma-spectrometry; single and correlation measurements. Methods to study the time behavior of nuclear states and interpretation of the results. Visual and electronic position sensitive detection. Particle identification techniques: deflection by electromagnetic fields, energy loss in materials and pulse shape analysis. Principles, construction and operation of high energy particle accelerators. Main ideas of experiments in neutron physics: sources, detectors and methods. Calorimetry in particle physics, oscillation. Principles and exploitation ways of nuclear energy. Applications of nuclear fission, nuclear safety. Thermonuclear fusion in the cosmos and the Earth: implementations and problems. Expected environmental impacts of the operation and fuel cycle of nuclear energetics. Comparison with other sources of energy.

# Literature

Compulsory:

G.F.Knoll: *Radiation Detection and Measurement*. (4th ed. J.Wiley, 2010.) *Recommended:* 

- 1) Relevant chapters in 2016 Review of Particle Physics, <u>Experimental Methods and</u> <u>Colliders</u>, <u>http://pdg.lbl.gov/2017/reviews/contents\_sports.html</u>
- 2) Lecture notes and some supplements: http://falcon.phys.unideb.hu/~raics/public/NukTech/English/

# Schedule:

1<sup>st</sup> week

Structure of the matter; interactions. Main physics quantities. Symmetries and conservation laws. General properties of nuclei, scheme of the nuclear interactions, excited states. Reaction kinematics.

 $2^{nd}$  week

Measurement procedures: spectroscopy, collisions (scattering), nuclear reactions. Results of the experiments: integral and differential cross sections, excitation function. Summary of the reaction models. Interaction of radiations with matter: detection and destruction.

 $3^{rd}$  week

General principles of the measurements. Components and properties of single channel experiments. Multi parameter systems: prompt relations and analysis of event-by-event data collections. Evaluation: decay statistics, error calculations, curve fitting. Counting losses. Sample preparation.

# $4^{th}$ week

Detector types and properties. Electronic signal processing. Differential and integral type measurements. Detection of charged particles. Gamma-spectrometry by scintillation and semiconductor detectors. Correlation methods. Activity determination.

# $5^{th}$ week

Time measurements; determination of the life time, half life. Electronic solutions for timing. Utilization of the nuclear physics principles (nuclear recoil, Doppler-shift, channeling effect, resonances, Mössbauer-effect).

# $6^{th}$ week

Particle identification methods:deflection in electric and magnetic fields, specific energy loss, pulse shape discrimination. Telescope technique. Methods for velocity measurements.

7<sup>th</sup> week

Visual methods for track detection: cloud and bubble chamber, light phenomena from gas discharge. Electronic methods for real-time processing: multiwire proportional and drift chambers, gas- and semiconductor strip detectors, time projection chamber, mosaic construction scintillation detectors, semiconductor matrix cameras.

8<sup>th</sup> week

Principles of high energy accelerators, construction and operation. Detection with electromagnetic and hadron calorimeters. System setup, operation, data handling and evaluation of the CERN Large Hadron Collider at its detector systems.

9<sup>th</sup> week

Instruments of neutron physics: sources, detectors, moderators. Methods of neutron spectrometry:proton scattering, time-of-flight, charged particle nuclear reactions, foil activation, slowing-down time and length.

# $10^{th}$ week

Physics of nuclear energetics: decay, fission, fusion. Operation principle, construction, types and generations of nuclear reactors. Implementation of the fuel cycles and extension their sustainability. Handling of nuclear waste and the environment. The Nuclear Power Plant in Paks: its role in the domestic electric energy supply; its further development.

# $11^{th}$ week

Nuclear reactons of the fusion process in the stars. Possible realization ways on the Earth with the heavy hydrogen izotopes. Operation principle of the Tokamak systems. Fuel cycle: production of deuterium and tritium. External and internal contamination. Technical problems in the realization of the fusion reactor.

# $12^{th}$ week

Role of the neutrinos in the weak ineraction. Indirect demonstration of their existence and the Debrecen experiment. Prompt and delayed methods of the direct demonstration. Diversity of the neutrinos in the Lepton-families. Oscillations.

# 13<sup>th</sup> week

Beginning of the neutrino astronomy. Observation of the Solar neutrino anomaly in the experiments and the solution to the problem. Observation of a supernova explosion by neutrinos and photons. Messengers and information of astronomical events by light, neutrinos, and gravitational waves.

# $14^{th}$ week

Utilization of nuclear methods in medical biology, agriculture, space technology and other fields. Significance of the nuclear interactions in the quantum physics. Summary and outlook. Consultation.

## **Requirements:**

#### - for a signature

The lectures are accompanied by students' seminars of general and special subjects. These are connected to the fundemental and applied research as well as to wide variety of technical applications. Students choose subjects at the beginning of the semester, mainly according to their interest.

#### - for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have a practical signature.

The grade for the examination is given according to the following table:

0-49       fail (1)         50-62       pass (2)         63-75       satisfactory (3)         76-88       good (4)         89-100       excellent (5)	Score		Grade
50-62pass (2)63-75satisfactory (3)76-88good (4)89-100excellent (5)	0-49		fail (1)
63-75satisfactory (3)76-88good (4)89-100excellent (5)	50-62		pass (2)
76-88good (4)89-100excellent (5)	63-75		satisfactory (3)
89-100 excellent (5)	76-88		good (4)
	89-100	)	excellent (5)

Person responsible for course: Dr. László Oláh, assistant professor, PhD

Lecturer: Dr. Péter Raics, associate professor, PhD

Title of course: Nuclear Astrophysics Code: TTFME163	<b>ECTS Credit points:</b> 3
Type of teaching, contact hours <ul> <li>lecture: 2 hours/week</li> <li>practice: -</li> <li>laboratory: -</li> </ul>	
Evaluation: exam	
Workload (estimated), divided into contact hours: - lecture: 28 hours - preparation for the exam: 40 hours Total: 68 hours	
Year, semester: 1 <sup>st</sup> year, 1 <sup>st</sup> semester	
Its prerequisite(s): -	
Further courses built on it: -	
Topics of course	
Elements of astronomy, observation technique, Hertzsprung-Ru nuclear physics, mass, binding energy, nucleus models, nuclea elements, isotope anomalies. Nucleosynthesis through the Big Bar Early stellar stages, hydrogen and helium burning. Solar models, of stellar evolution, supernovae, neutron stars, black holes. Heavy process, s-process, p-process. Nuclear reaction types, reaction methods to determine reaction cross sections. Low background stu Cosmic rays, dark matter, dark energy. Cosmochronology, galacti	ussel diagram. Elements of r reactions. Abundances of ng and in the early universe. solar neutrinos. Final stages element nucleosynthesis, r- on networks. Experimental dies. Exotic nuclear physics. c chemical evolution.
Literature	
C.E. Rolfs, W.S. Rodney: Cauldrons in the Cosmos, University of Chic C. Iliadis: Nuclear Physics of Stars, WILEY-VCH, 2008	cago Press, 1988.
Schedule: $I^{st}$ week Elements of astronomy, observation technique, Hertzsprung-Russ $2^{nd}$ week Elements of nuclear physics, mass, binding energy, nucleus mode $3^{rd}$ week Abundances of elements, isotope anomalies. $4^{th}$ week Nucleosynthesis through the Big Bang and in the early universe. $5^{th}$ week Early stellar stages, hydrogen and helium burning. $6^{th}$ week Solar models, solar neutrinos. $7^{th}$ week Final stages of stellar evolution, supernovae, neutron stars, black I $8^{th}$ week	el diagram. ls, nuclear reactions. holes.

Heavy element nucleosynthe	esis, r-process, s-process, p-process.	
9 <sup>th</sup> week		
Nuclear reaction types, react	ion networks.	
10 <sup>th</sup> week		
Experimental methods to det	ermine reaction cross sections.	
11 <sup>th</sup> week		
Low background studies.		
12 <sup>th</sup> week		
Exotic nuclear physics.		
13 <sup>th</sup> week		
Cosmic rays, dark matter, da	rk energy.	
14 <sup>th</sup> week		
Cosmochronology, galactic of	chemical evolution.	
<b>Requirements:</b>		
- for a signature		
Attendance at lectures is comp	sulsory. A student must attend the lectures and may not miss more than	
three times during the semester. In case a student does so, the subject will not be signed and the student		
must repeat the course.		
- for a grade		
The course ends in an <b>examination</b> . The requirement for applying for an exam is to have a practical		
signature.		
The grade for the examination	is given according to the following table:	
Score	Grade	
0-49	fail (1)	
50-62	pass (2)	
63-75	satisfactory (3)	
76-88	good (4)	
89-100	excellent (5)	
Person responsible for course: Zsolt Fülöp, PhD		
Lecturer: Zsolt Fülöp, PhD, G	ábor Kiss, PhD	

Title of course: Nuclear physics laboratory Code: TTFML0164	ECTS Credit points: 4
Type of teaching, contact hours	
- lecture: -	
- practice: -	
- laboratory: 4 hours / week	
Evaluation: lab report and oral presentation	
Workload (estimated), divided into contact hours:	
- lecture: -	
- practice: -	
- laboratory: 52 hours	
- home assignment: 50 hours	
- preparation for the exam: 10	
Total: 112 hours	
<b>Year, semester</b> : 1 <sup>st</sup> year, 2 <sup>nd</sup> semester	
Its prerequisite(s): -	
Further courses built on it: -	
Topics of course	
performed using HPGe detectors. Study the basic steps of a gamma s identifying radioisotopes in samples of unknown origin, carry out gam spectrum evaluations. Determination of the gamma photon energy efficiency of a HPGe detector when simple detection geometry is used the detector, practice the way of measurement and calculation of the unknown origin.	pectroscopy procedure used for ma spectrum measurements and dependence of the detection (point like sample is counted by gamma activity of a sample of
Literature	
György Gyürky – Energy calibration of the accelarator with Al(p,g)Si András Fenyvesi – Gamma Spectroscopy Measurements with High Pur András Fenyvesi – Measurement of the cross section of the 56Fe(n,p) neutron energy via the activation method	reaction ity Germanium Detector System 56Mn reaction at En = 9.9 MeV
Schedule:	
1 <sup>st</sup> week	
Al target preparation	
$2^{nd}$ week	
Preparation of the stack to be irradiated	
3 <sup>rd</sup> week	
Energy calibration of the accelarator by measuring the reaction of 27A	l(p,γ) 28Si
4 <sup>th</sup> week	
Preparation of the stack to be irradiated	
Activation of samples with quasi-monoenergetic neutrons (irradiation o	f the prepared aluminum and iron
discs).	
6 <sup>th</sup> week	

Counting of the activated samples with the HPGe spectrometers

 $7^{th}$  week

Evaluation of the full energy gamma peaks in the spectra using the programs of the mul-tichannel analyzers (MCAs).

 $8^{th}$  week

Calculation of the neutron flux and the cross section

9<sup>th</sup> week

Comparison of the obtained cross section with data of evaluations. Comparison and dis-cussion of the results of the three groups. Discussing uncertainty and error of the meas-urement.

 $10^{th}$  week

Energy calibration of the HPGe detector

11<sup>th</sup> week

Evaluation of the experimental results and fabrication of the report.

 $12^{th}$  week

The presentation of the report of the experimental results.

13<sup>th</sup> week

Optional consultations

14<sup>th</sup> week

Catch up laboratory work

# **Requirements:**

- for a signature

Signature requires the attendance at all laboratory works.

- for a grade

Knowledge of definitions, laws and theorems: grade 2;

In addition, knowledge of the proofs of most important theorems: grade 3;

In addition, knowledge of the proofs of theorems: grade 4;

In addition, knowledge of applications: grade 5.

Person responsible for course: Dr. Sándor Nagy, associate professor, PhD

Lecturer: Dr. Balazs Ujvari, assistant professor, PhD

<b>Title of course: Insight into the Present Day Nuclear Research</b> <b>Code</b> : TTFME0165	ECTS Credit points: 4
Type of teaching, contact hours	
- lecture: 2 hours/week	
- practice: 1 hours/week	
- laboratory: -	
Evaluation: exam	
Workload (estimated), divided into contact hours:	
- lecture: 28 hours	
- practice: 14 hours	
- laboratory: -	
- home assignment: 40 hours	
- preparation for the exam: 40 hours	
Total: 122 hours	
<b>Year, semester</b> : 1 <sup>st</sup> year, 2 <sup>st</sup> semester	
Its prerequisite(s): -	
Further courses built on it: -	
Topics of course	
Instruments and scientific applications of precision atomic mass measur of the constancy of the radioactive decay constant. Nucleon beam and q	rement. Experimental verification uantum electrodynamics. Current

of the constancy of the radioactive decay constant. Nucleon beam and quantum electrodynamics. Current state of energy-producing fusion experiments. The nuclear physics "pandemium" effect and its role in the heat production of nuclear reactors. Radioactive ion beams. Experimental study of exotic nuclei. Measuring the cross section of nuclear reactions through the example of a specific reaction. Nuclear astrophysical experiments in a deep underground laboratory. Special modes of rotation in nuclei. Nuclear power - yesterday, today, tomorrow. Methods of evaluation of gamma spectra. Gamma spectroscopic studies on radioactive beams.

Topics may change depending on new findings in scientific research.

# Literature

Depending on the topics covered during the semester, the literature recommended by the speaker.

#### Schedule:

1<sup>st</sup> week

Instruments and scientific applications of precision atomic mass measurement. The working principle of the Penning trap. Applications of high precision data. Experimental control of QED (measurement of free and bound electron g-factor). CPT symmetry test (g-factor of electron and positron; measurement of proton-antiproton mass difference). Determination of basic physical constants (fine structure constant; mp/me ratio). Measurement of the 3H-3He mass difference for the KATRIN program to determine the neutrino mass. Experimental verification of the CKM matrix. Preparing for the introduction of a new kilogram standard.

 $2^{nd}$  week

Experimental verification of the constancy of the radioactive decay constant. The unprecedented accuracy of the study of the constants of various decomposition processes through the years of precision

work organized by a network of large international metrology laboratories. Exclusion of solar gravitational field influence.

# 3<sup>rd</sup> week

Nucleon beam and quantum electrodynamics. Search for finite proton size and first experimental proof of QED (Lamb). Questioning the equal QED manageability of electron and muon (Pohl). Current experimental attempts and future plans to solve the "mystery" of the proton and deuteron rays.

# $4^{th}$ week

Current state of energy-producing fusion experiments. General overview. JET, the largest tokamak through DT operation. ITER, the first fusion reactor. The future of fusion energy production. A new initiative: the Wendelstein 7-X Stellarator.

# $5^{th}$ week

The nuclear physics "pandemium" effect and its role in the heat production of nuclear reactors.

Methods of analysis of the  $\gamma$ -spectrum of fission products; collision of energy resolution and efficiency. A systematic source of error in level construction: the nuclear physics "pandemonium" effect. Corollary: Uncertainty in the calculation of heat of decomposition. Advantages of using a full absorption gamma (TAG) spectrometer.

 $6^{th}$  week

Radioactive ion beams. Methods for generating radioactive ion beams used to study the properties of nuclei far from the stability band; the methods for identifying the particles making up the beam, the detectors used in typical measurements and their basic characteristics.

# $7^{th}$ week

Experimental study of exotic nuclei. Methods for measuring the special distribution of neutrons and protons in the nucleus. Neutron skin, neutron halo and their astrophysical significance

 $8^{th}$  week

Measuring the cross section of nuclear reactions through the example of a specific reaction.

Demonstration of the steps required for the experimental determination of the cross-section of an effect through the example of a specific nuclear reaction, from the characterization of the targets through the irradiations with the accelerator to the detection of the reaction products.

9<sup>th</sup> week

Nuclear astrophysical experiments in a deep underground laboratory. Special techniques that are required to measure very low impact cross sections. Low background radiation in a deep underground laboratory is advantageous for the measurements. Demonstration of the characteristics of nuclear astrophysical research with an underground accelerator through the example of LUNA's international cooperation, which is unique in the world

 $10^{th}$  week

Special modes of rotation in nuclei.

Description of the characteristic collective form of motion of deformed (non-spherical) nuclei, rotation. Special phenomena resulting from the interaction of the internal state and rotation of the nucleus: e.g. chiral rotation, oscillating rotation, closure of rotation bands.

11<sup>th</sup> week

Nuclear power - yesterday, today, tomorrow.

Energy conditions of nuclear physics processes and possibilities of their use. Nuclear fission as a source of energy. Self-sustaining chain reaction. Internal (physical) and external (technical) security. Reactor types and their development directions. Fuel cycle. Energy from fusion. Hybrid solutions. Radioactive waste. Social attitude, politics.

 $12^{th}$  week

Methods of evaluation of gamma spectra. Analysis of multidimensional coincidence gamma matrices. Structure of the system of excited states (level scheme). Determining the multipolarity of gamma transitions using angle correlation analysis; determination of quantum mechanical characteristics (spin and parity values) of excited states based on the obtained multipolarities

# 13<sup>th</sup> week

Gamma spectroscopic studies on radioactive beams.

Obtaining basic information about the structure of exotic nuclei, the properties of the excited states of nuclei by gamma spectroscopic evaluation of experiments with radioactive beams. Characterization of the latest gamma detectors.

*14<sup>th</sup> week* Consultation.

#### **Requirements:**

- for a signature

Attendance at **lectures** and participation at **practice classes** is compulsory. A student must attend the classes and may not miss more than three times during the semester. In case a student does so, the subject will not be signed and the student must repeat the course.

-for a grade

Every students have to give two presentations on topics chosen from the material of the semester and to answer to some basic questions. The grade is based on this performance.

Person responsible for course: Dr. Judit Darai, associate professor, PhD

Lecturer: Dr. Judit Darai, associate professor, PhD

Title of course: Condensed Matter Physics 4	ECTS Credit points: 5
Code: TTFME0172	
Type of teaching, contact hours	
- lecture: 2 hours/week	
- practice: 1 hours/week	
- laboratory: -	
Evaluation: exam	
Workload (estimated), divided into contact hours:	
- lecture: 28 hours	
- practice: 14 hours	
- laboratory: -	
- home assignment: 40 hours	
- preparation for the exam: 40 hours	
Total: 122 hours	
<b>Year, semester</b> : 1 <sup>st</sup> year, 1 <sup>st</sup> semester	
Its prerequisite(s): -	
Further courses built on it: -	

#### **Topics of course**

Drude model (basic equation of motion, metallic shine, electric and thermal conductivity, Lorentz number, plasma frequency, Hall effect, Seebeck effect, model evaluation based on experimental results); Non-interacting quantum mechanical electron system at T=0 and non-zero temperatures, Fermi surface, compressibility, Sommerfeld model (Sommerfeld expansion, characterization of different physical quantities of interest, study of the specific heat, density of states, effective mass, model evaluation based on experimental data); Phonons, Fermi liquids, non-Fermi liquids, lattice diffraction, Bloch electron. The effect of week periodic potential (non-degenerate and degenerate cases), Tight-binding method, band insulator, Mott insulator, Wigner lattice, semiclassical model, hole notion. Landau levels, quantum Hall effect, topological insulator, spin quantum Hall effect, anomalous quantum Hall effect, heavy-fermion systems, band topology.

#### Literature

Compulsory: N. W. Ashcroft, N. D. Mermin: Solid State Physics, Saunders College Publishing, 1976, US Recommended: P. W. Anderson: Concepts in Solids, Benjamin/Cummings Publishing Company, London 1982.

#### Schedule:

1<sup>st</sup> week

Model starting points for the Drude model, appearance circumstances, the relaxation

time notion, treatement of collision process, deduction of the model equation of motion and its interpretation.

 $2^{nd}$  week

The results provided by the Drude model in the presence of an external electric

field constant in r-space but periodic in time. The electrical conductance of the Drude model and its characteristics. The Drude resistivity in the light of experimental data.  $3^{rd}$  week The results provided by the Drude model in the presence of an external electric field periodic both in space and time. Metallic shine, plasma frequency and its physical interpretation and measurement. Comparison to experimental data. The introduced concepts seen today: the plasmon notion..  $4^{th}$  week

The Hall effect seen in the frame of the Drude model. Evaluation of the theoretical result and comparison to experimental data. The Hall effect seen today: Landau levels, the quantum-Hall effect, the spin quantum-Hall effect, the anomalous quantum-Hall effect, topological insulators and their importance.  $5^{th}$  week

The thermal conductivity in the frame of the Drude model. Comparison to thermal conductivity deduced for gases. The Lorentz number, the Wiedemann-Franz low, deduction of the Lorentz number in the frame of the Drude model, comparison to experimental data. The importance of the Lorentz number in physical processes today.

# $6^{th}$ week

Seebeck effect, thermopower and its deduction in the frame of the Drude model, physical interpretation of the deduced result and comparison to experimental data. Evaluation of the Drude model, failures of the model and limits of the Drude model. Study of the

development possibilities of the model description. The importance of the Seebeck effect in the modern condensed matter physics.

 $7^{th}$  week

Quantum mechanic non-interacting electron gas at T=0 temperature. Born-von Karman periodic boundary conditions, possible states in k-space, summation possibilities over the momentum value, the notion of the density of states, its calculation technique, and its expression for non-interacting Fermi gas. Fermi surface, density of states on the Fermi surface, Fermi energy, Fermi velocity, Fermi temperature, comparison to experimental data.

 $8^{th}$  week

Ground state energy at T=0, calculation of the internal pressure, deduction of the equation of state, comparison to the properties of the non-interacting classical gas. Deduction of the compressibility, its evaluation, and comparison to experimental data taken in  $T \rightarrow 0$  limit.

Role of the Fermi gas compressibility in the rigidity of solid state matter.

9<sup>th</sup> week

Non-interacting quantum Fermi gas at non-zero temperatures. Partition function, thermodynamic potential, deduction of the relation for internal energy, entropy and total number of particles. The Fermi function and its approximation possibilities. The momentum distribution function and its discussion at current knowledge level: the effect of the interaction, importance of cusp points.

 $10^{th}$  week

The Sommerfeld expansion, the Sommerfeld model and its results at low non-zero temperatures. The calculation of the chemical potential, and its relation to the Fermi energy. Calculation of the specific heat, and its connection to the effective mass. Comparison to the results deduced in the Drude model, and experimental data. The evaluation of the Sommerfeld model, its limits, study of the development possibilities of the model description. Importance of the introduced concepts in the current condensed matter physics: Fermi liquid notion, Non-Fermi liquids, known rigorous non-Fermi liquid in 1D: Luttinger liquid, short presentation of heavy fermion systems.

11<sup>th</sup> week

Point lattice, Bravais lattice, Voronoy construction and polyhedron, Lattice plain families and their characteristics, reciproc lattice and its characteristics in D=1,2,3 dimensions, connection between reciproc lattice vectors and lattice plain families, Miller vectors.

Bravais lattices constructed based on their cells, characteristic cells and their properties, the zone notion, and its properties. Phonons

 $12^{th}$  week

Waves propagation in a lattice, diffraction caused by the lattice, Bragg and von Laue treatement of the diffraction, reciproc space, and Brillouin zone notions. Experimental definition of the reciproc lattice and Brillouin zone boundaries. Form factors and their role. Bloch theorem in the light of symmetry properties, Bloch theorem in the light of the Schrödinger equation, the notion of Bloch electron.

# $13^{th}$ week

Fourier transformation in the Bravais lattice, Schrödinger equation in the independent electron approximation. The action of a small periodic potential: non-degenerate and degenerate cases. The behavior of the electron in the center of the Brillouin zone and close to the Brillouin zone boundary. The gap caused by the periodic potential, the band notion, movement inside the band, conducting and insulating systems, umklapp processes and their importance, the Johns zone notion and its importance.  $14^{th}$  week

Wannier functions tight binding approximation and the dispersion relation that it provides, Mott insulator, Wigner lattice, the Semiclassical model, its characteristics and limits, the hole notion and its importance. The condensed matter physics seen today, strongly

correlated systems.

#### **Requirements:**

- for a signature

Attendance at **lectures** is recommended, but not compulsory.

Participation at **practice classes** is compulsory. A student must attend the practice classes and may not miss more than three times during the semester. In case a student does so, the subject will not be signed and the student must repeat the course. During the semester there will be two written examinations (on 6th and 13th week) from lectures and practice as well. The requirement for a signature is a successful (> 20%) completion of each written exam.

- for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have a practical signature.

The grade for the examination is given according to the following table:

Score	Grade
0-49	fail (1)
50-62	pass (2)
63-75	satisfactory (3)
76-88	good (4)
89-100	excellent (5)
Person responsible for course: Dr. Zsolt Gulácsi, professor, DSc	
Lecturer: Dr. Zsolt Gulácsi, professor, DSc	

Title of course: Quantum Mechanical Many-Body Physics 1	ECTS Credit points: 5
Code: TTFME0174	-
Type of teaching, contact hours	
- lecture: 2 hours/week	
- practice: 1 hours/week	
- laboratory: -	
Evaluation: exam	
Workload (estimated), divided into contact hours:	
- lecture: 28 hours	
- practice: 14 hours	
- laboratory: -	
- home assignment: 40 hours	
- preparation for the exam: 40 hours	
Total: 122 hours	
<b>Year, semester</b> : 1 <sup>st</sup> year, 1 <sup>st</sup> semester	
Its prerequisite(s): -	
Further courses built on it: -	

#### **Topics of course**

Second quantization, S matrix, Gell-Mann Low theorem, P and T products, expectation value of time ordered operator products. T=0 description: One particle Green-function, expectation value of different Hamiltonian terms, ground state energy, Gorkov equation, one-particle propagators for non-interacting Fermi and Bose systems, Lehmann representation, spectral functions, Wick theorem and its application, Feynmann diagrams, Dyson equation, self-energy contributions, polarization insertion, two-particle Green function, Vertex function, Bethe-Salpeter equation. Non-zero T description: The Matsubara

technique, T > 0 Green functions, summation over Matsubara frequencies in fermionic and bosonic cases, calculation of expectation values at T > 0 in terms of Green-functions, the relation between the thermodynamic potential and T > 0 Green function, Feynman diagram technique at non-zero temperatures.

#### Literature

Compulsory: A. L. Fetter, J. D. Walecka: Quantum theory of many particle systems, McGraw-Hill Book Company, New York, 1975 Recommended: A. A. Abrikosov, L. P. Gorkov, I. Y. Dzyaloshinskii: Quantum Theoretical Methods in statistical Physics, Dover Publication Inc., New York, 1975.

## Schedule:

1<sup>st</sup> week

Second quantization in Fermi and Bose cases, the relation between first and second quantization.  $2^{nd}$  week

Quantum field theory methods used at T=0 temperature. Quantum mechanical pictures and the connection in between them. S matrix: definition, expression, importance.

 $3^{rd}$  week

Adiabatic swithing on, the Gell-Mann Low theorem: deduction, interpretation, importance.

.4<sup>th</sup> week

T and P products, calculation of expectation values from T products, the use of the S matrix in this process, the propagator notion.

 $5^{th}$  week

The one-particle Green function at T=0 temperature. The Gorkov equation, deduction and use.  $6^{th}$  week

Deduction of the expectation values of different operators in terms of the one-particle Green function in the fermionic and bosonic case.

 $7^{th}$  week

Analitic properties of one-particle Green functions, spectral functions and Lehmann representation..

 $8^{th}$  week

*The poles of the Green function, connection between T and P roducts, Wick theorem.* 

 $9^{th}$  week

Feynmann diagrams, diagram expansion of the expression of the one-particle Green function, self energy contribution, Dayson equation and its importance.

 $10^{th}$  week

*Green functions in k-space, Feynmann diagrams in k-space, notion of the polarization loop.* 11<sup>th</sup> week

*Two-particle Green functions, notion of the vertex function, Bethe-Salpeter equation.* 

 $12^{th}$  week

One-particle Green function at non-zero temperatures, and its properties. Feynmann diagrams at non-zero temperatures.

 $13^{th}$  week

Matsubara frequencies, the Green function expressed as a sum over Matsubara frequencies, the technique of calculation of the sums over Matsubara frequencies in fermionic and bosonic cases.

 $14^{th}$  week

Deduction of the thermodynamic potential from the one-particle Green function at non-zero temperatures, calculation of thermodynamic averages. The calculation of the change in the thermodynamic potential when a new interacting term emerges in the Hamiltonian.

# **Requirements:**

- for a signature

Attendance at **lectures** is recommended, but not compulsory.

Participation at **practice classes** is compulsory. A student must attend the practice classes and may not miss more than three times during the semester. In case a student does so, the subject will not be signed and the student must repeat the course. During the semester there will be two written examinations (on 6th and 13th week) from lectures and practice as well. The requirement for a signature is a successful (> 20%) completion of each written exam.

- for a grade

The course ends in an **examination**. The requirement for applying for an exam is to have a practical signature.

The grade for the examination is given according to the following table:

Score	Grade
0-49	fail (1)
50-62	pass (2)
63-75	satisfactory (3)
76-88	good (4)
89-100	excellent (5)
Person responsible for course: Dr. Zsolt Gulácsi, professor, DSc	
Lecturer: Dr. Zsolt Gulácsi, professor, DSc	

<b>Title of course</b> : Basics of Functional Renormalization Group Method <b>Code</b> : TTFME0175	ECTS Credit points: 5
Type of teaching, contact hours	
- lecture: 2 hours/week	
- practice: 1 hours/week	
- laboratory: -	
Evaluation: exam	
Workload (estimated), divided into contact hours:	
- lecture: 28 hours	
- practice: 14 hours	
- laboratory: -	
- home assignment: 14 hours	
- preparation for the exam: 28 hours	
Total: 84 hours	
<b>Year, semester</b> : 2 <sup>nd</sup> year, 2 <sup>nd</sup> semester	
Its prerequisite(s): -	
Further courses built on it: -	
Topics of course	
Introduction of the sine-Gordon (SG) model as a discrete system of torsional harmonic oscillators. Symmetries and phase structure of SG type models. The bosonisation and SG type models. Topological phase transitions of SG type models. SG models in higher dimensions and their relation to Higgs, inflaton	

and axion physics. The Kadanoff-Wilson blocking construction. The Wegner-Houghton, the Polchinski and the Wetterich functional renormalization group (FRG) equations. Gradient expansion and regulator dependence. Linearised FRG equations for SG type models in the leading order of the gradient expansion, i.e., in the local potential approximation (LPA). Exact FRG equations for SG type models with a single frequency in LPA. Exact FRG equations for multi-frequency SG type models in LPA. Beyond LPA, the derivation and the solution of the linearised FRG equations for SG type models. Beyond LPA, the derivation and the solution of the exact FRG equations for SG type models with single and multi frequencies.

# Literature

Compulsory:

Nándori István, Lecture Note on the Functional Renormalization Group Study of Sine-Gordon Models, electronic lecture note.

Recommended:

Nagy Sándor, Lectures on renormalization and asymptotic safety, Annals Phys. 350 (2014) 310-346, Sailer Kornél, Renormálási csoport a kvantumtérelméletben, KLTE 1997, Sailer Kornél, Szimmetriák II., KLTE 1994-2013.

# Schedule:

1<sup>st</sup> week

The classical (relativistic but not quantised) sine-Gordon model is introduced by taking the continuous limit of a discrete system of torsional harmonic oscillators. A brief introductions to QFT, renormalization and renormalization group are also peresented.

 $2^{nd}$  week
Periodic (sine-Gordon type) models are overviewed and the consequence of their symmetries on their phase structure is discussed.

 $3^{rd}$  week

The bosonisation transformations, i.e., mappings of two-dimensional fermionic and gauge theories such as the multi-color QCD\_2, and the multi-flavour QED\_2, onto bosonic models are summarised and their connetcions to sine-Gordon type models are given.

 $4^{th}$  week

Topological phase transitions are discussed by rewriting the generating functional of sine-Gordon theories onto partition functions while the path-intgeral is performed. Their relation to classical spin models and the O(2) symmetric scalar theory and the vortex dynamics of superconducting and superfluid systems are presented.

 $5^{th}$  week

Applications of sine-Gordon models in higher dimensions and in Conformal Field Theory. The relation of sine-Gordon models to Higgs, inflaton and axion physics is discussed.

6<sup>th</sup> week

The Kadanoff-Wilson blocking construction is discussed which is based on scale-invariance of the system undergoes a second order phase transition. It is shown that the Wegner-Houghton and the Polchinski RG equations can be derived based on the blocking construction by using the idea of Kenneth G. Wilson.

7<sup>th</sup> week

A brief introduction to the modern form of the FRG apporach, i.e., the derivation of the Wetterich RG equation is presented and its relation to the Wegner-Houghton and Polchinski RG equations is given. Important technical issues, such as gradient expansion, regulator dependence etc. are also discussed.

 $8^{th}$  week

The derivation of the linearised FRG equations (for sine-Gordon type models) obtained in the leading order of the gradient expansion (LPA) is presented. Discussion of the phase structure of sine-Gordon type models based on the solution of these RG equations is also given.

9<sup>th</sup> week

Exact FRG equations for sine-Gordon type models with a single frequency are presented and their solutions are studied.

10<sup>th</sup> week

The FRG study of multi-frequency sine-Gordon theories are presented in LPA. The connection of the RG results to the requirement of convexity is shown.

11<sup>th</sup> week

Beyond LPA, the derivation and the solution of the linearised FRG equations for sine-Gordon

type models are shown.

 $12^{th}$  week

Beyond LPA, the derivation and the solution of the exact FRG equations for sine-Gordon

type models with single and multi frequencies are presented.

13<sup>th</sup> week

Results obtained by the FRG method are summarised.

14<sup>th</sup> week

Consultation.

## **Requirements:**

- for a signature

Attendance at **lectures** is recommended, but not compulsory.

- for a grade

The course ends in an <b>examination</b> .	
The grade for the examination is given according to the following table:	
Score	Grade
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89-100	excellent (5)
Person responsible for course: Dr. István Nándori, associate professor, PhD	
Lecturer: Dr. István Nándori, associate professor, PhD	